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**Research Article** 

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# Synthesis of deuterium labelled chloroquine, hydroxychloroquine and their metabolites

Lei Tian<sup>1</sup>, Chi Zhang<sup>2</sup> and Jian Li<sup>3\*</sup>

<sup>1</sup>College of Petroleum Engineering, Yangtze University, 1 Daxue road, Caidian Zone, Wuhan, Hubei, China <sup>2</sup>School of Biochemical and Environmental Engineering, Nanjing Xiaozhuang University, 3601 Hongjing Road, Nanjing, Jiangsu, China

<sup>3</sup>Hi-Tech Research Institute and State Key Laboratory of Materials-Oriented Chemical Engineering, Nanjing University of Technology, 5 Xinmofan Road, Nanjing, Jiangsu, China

# ABSTRACT

This paper describes the synthesis of deuterium-labelled chloroqiune, hydroxychloroquine and their metabolites. Mass spectrometry analysis of these compounds revealed over 98% deuterium enrichment.

Keywords: deuterium-labelled; chloroquine; hydroxychloroquine; metabolite; synthesis

# INTRODUCTION

The popularity of chloroquine and hydroxychloroquine for malaria treatment in many Third World countries emanates from it being cheap, widely available, relatively well tolerated, and having a rapid onset of action [1-4]. Besides being active against malaria, they are used to treat rheumatoid arthritis [5-6] and cutaneous lupus erythematosus (LE) and rashes associated with systemic lupus erythematosus (SLE) [7-8]. They are also used in some photosensitivity disorders and occasionally in other inflammatory skin conditions [9-10]. In addition, chloroquine and hydroxychloroquine are two inexpensive agents that have been shown to achieve some level of anti-HIV activity [11]. Metabolism studies revealed that chloroquine and hydroxychloroquine are mainly metabolized to the still active metabolite desethylchloroquine, as shown in Figure 1 [12-13]. Although <sup>3</sup>H&<sup>14</sup>C chloroquine and hydroxychloroquine have been prepared for pharmacological studies, the synthesis of their stable labeled internal standard has not been described in details. In this paper, the synthetic route to [<sup>2</sup>H<sub>4</sub>] chloroquine, [<sup>2</sup>H<sub>5</sub>] hydroxychloroquine and their metabolites were described in detail.

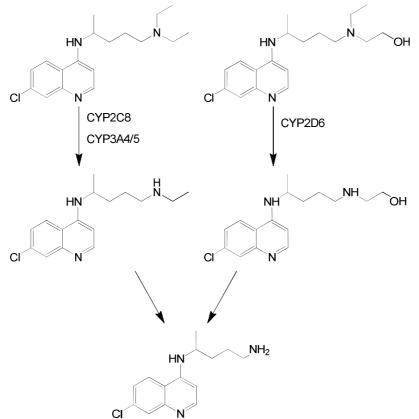
#### **EXPERIMENTAL SECTION**

#### General

All reagents were obtained from Sigma-Aldrich and CDN Isotope. Mass spectra were recorded using a Quattro micro API mass spectrometer. <sup>1</sup>H NMR spectra were recorded on a Bruker 300 MHz instrument (Bruker Corporation, Germany). Chemical purities were determined by an Agilent 1200 HPLC with a XDB-C18 column, 5  $\mu$ m, 4.6 mm×150 mm (Agilent, USA).

#### Synthesis of (Z)-5-hydroxypentan-2-one oxime (2)

A mixture of 5-hydroxypenta-2-one (1) (80 g, 0.78 mol) and hydroxylamine (65 g, 0.94 mol) in water (640 mL) was added KOH (37 g, 0.67 mol) slowly under ice/water bath. The reaction solution was stirred for 8 h at room temperature under  $N_2$ . The reaction solution was cooled to 0 °C and acidified to pH 8 by adding conc. HCl. The solution was extracted by EtOAc (200 mL×8). The organic layers were combined, dried by  $Na_2SO_4$  and



concentrated under reduced pressure to give (2) as a light yellow oil (41.36 g, 90%).

Figure 1. The metabolite routes of chloroquine and hydroxychloroquine

#### Synthesis of 4-aminopentan-1-ol (3)

A slurry of (2) (15 g, 0.13 mol) and Raney Ni (5 g) in MeOH (60 mL) was placed in a stainless steel pressure vessel, which was pressurized to 0.8 Mpa with  $H_2$  gas. The reaction was mechanically stirred for 4 h at 65 °C. The mixture was cooled to room temperature. The Raney Ni was removed by filtration through celite, and the solid was rinsed with MeOH (25 mL×3). The combined filtrates were concentrated to dryness to give (3) as a little blue oil (13.1 g, 98%).

<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ3.66 (t, 2H, J= 7.1 Hz), 2.89 (m, 1H), 2.81 (brs, 3H), 1.69 (m, 3H), 1.35 (m, 1H), 1.12 (d, 3H, J=6.8 Hz).

#### Synthesis of 4-(7-chloroquinolin-4-ylamino)pentan-1-ol (4)

To a solution of compound (3) (27 g, 0.262 mol) and 4, 7-dichloroquinoline (25.9 g, 0.131 mol) was stirred at 140 °C for 5 h. The reaction mixture was cooled to 100°C, poured into water (100 mL) and stirred to induce crystallization. The resulting solid was collected by filtering and rinsed thoroughly with water (50 mL×3). The crude product was recrystallized from methanol/ethanol to give (4) as colorless solid (16.8 g, 48.41%).

<sup>1</sup>H NMR (DMSO, 300 Hz): δ 8.36 (d, 1H, J=7.5 Hz), 8.34 (d, 1H, J= 1.5 Hz), 7.75 (d, 1H, J= 7.5 Hz), 7.40(d, 1H, J=7.5 Hz), 6.93 (d, 1H, J=7.5 Hz), 6.42 (d, 1H, J=2.5 Hz), 4.56 (brs, 1H), 1.45-1.85 (m, 4H), 1.19 (t, 3H, J=6.8 Hz).

#### *Synthesis of N-(5-bromopentan-2-yl)-7-chloroquinolin-4-amine hydrobromide (5)*

To a solution of 48% hydrobromic acid (45.5 mL) was added conc.  $H_2SO_4$  (9.7 mL). Compound (4) (13 g, 49.1 mmol) was dissolved in the acid solution, and the resulting solution heated to 105°C as rapidly as possible. The mixture was stirred continuously for 15 min. The solution was cooled to room temperature and diluted with water (50 mL). The solution was extracted with  $CH_2Cl_2$  (50 mL×6). The combined organic layer was dried over  $Na_2SO_4$  and concentrated under reduced pressure to give (5) as a grey solid (16.5 g, 82.3%).

<sup>1</sup>H NMR (DMSO, 300 Hz): δ 1.19 (3H, t, J= 6.8 Hz), 1.43-1.68 (4H, m), 1.45-1.85 (4H, m), 3.42 (2H, t, J=7.1 Hz), 4.56 (1H, brs), 6.42 (1H, br d, J= 5.4 Hz), 6.93 (1H, d,), 7.40(1H, d), 7.76 (1H, d), 8.52 (2H, dd),

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# Synthesis of $N^4$ -(7-chloroquinolin-4-yl)- $N^1$ -ethylpentane-1, 4-diamine (6)

To a solution of (5) (11g, 26.92 mmol) in methanol (100 mL) was added ethylamine solution (70%, 110 mL) under ice/water bath. The solution was stirred for 8 h at room temperature. TLC showed little starting material remained. The solution was co-evaporated with ethanol (150 mL×3) to afford a yellow solid. The solid was purified by column chromatography on silica gel column, eluted with  $CH_2Cl_2$ /saturated methanol ammonia (10: 0.3) to afford (6) as a silver gray solid (3.62 g, 46.07%).

# *Synthesis of* $[{}^{2}H_{5}]$ *chloroquine diphosphate (7)*

To a suspension of (6) (1.8 g, 6.17 mmol) and  $K_2CO_3$  (0.85 g, 6.17 mmol) in dry DMF (18 mL) was added [<sup>2</sup>H<sub>5</sub>] ethyl iodide slowly. The reaction solution was stirred for 8 h at room temperature. The mixture was diluted with water (150 mL) under water/ice bath and extracted with EtOAc (100 mL×9). The combined organic layers were evaporated to dryness to afford a colorless liquid. The crude product was purified by column chromatography on silica gel column, eluted with CH<sub>2</sub>Cl<sub>2</sub>/ saturated methanol ammonia (10: 0.7) to afford a silver gray solid (3.62 g, 46.07%). The solid (0.5 g, 1.54 mmol) was dissolved in dry ethanol (4 mL) and heated under reflux. The solution was adjusted to pH 5 by adding H<sub>3</sub>PO<sub>4</sub> (0.36 g, 85%). The mixture was stirred continuously for 3 h. The resulting solid was collected by filtering, and compound (10) was obtained as a white solid (0.651g, 81.2%).

<sup>1</sup>H NMR (DMSO) δ 8.38 (dd, 2H, J= 7.5, 7.5 Hz), 7.74 (d, 1H, J=2.1 Hz), 7.42(d, 1H, J=9.0 Hz), 7.02 (d, 1H, J= 7.5 Hz), 6.50 (brd, 1H, J= 5.4 Hz), 2.35-2.75 (m, 5H), 1.45-1.85 (m, 4H), 1.22 (d, 3H, J= 6.3 Hz), 1.02 (t, 3H, J= 6.9 Hz). MS-EI, (m/z): 325.2 (100), 326.2 (22), 327.2 (36), 328.2 (8). HPLC (XDB-C18, wavelength= 240 nm, CH<sub>3</sub>OH/10mmol/L K<sub>2</sub>HPO<sub>4</sub>=83/17, 1.0 mL/min): t<sub>R</sub> 6.79 min (98.7%). Isotopic enrichment determined by MS was over 98%.

# *Synthesis of* $[{}^{2}H_{4}]$ *2-(ethylamino)ethanol (9)*

To a stirred solution of  $[^{2}H_{4}]$ 2-bromoethanol (8) (2 g, 15.51 mmol), CH<sub>3</sub>CH<sub>2</sub>NH<sub>2</sub> (5.59 g, 124.04 mmol) in EtOH/H<sub>2</sub>O (25 mL) was heated at 85°C for 8 h. The solution was cooled by ice/water bath and acidified by conc. HCl to pH 2. The mixture was co-evaporated by EtOH to remove water out. White solid and oily solution was appeared. Na<sub>2</sub>CO<sub>3</sub> solid (15 g) and ether (100 mL) were added to solution to adjust pH=10. The organic layer was separated, and the aqueous layer was extracted with ether (100 mL×5). The combined organic phases were concentrated under reduced pressure to give (9) as a colorless liquid (1.44 g, 28.1%).

# Synthesis of $[{}^{2}H_{4}]$ hydroxychloroquine sulfate (10)

A mixture of compound (5) (0.85 g, 2.59 mmol) and compound (9) (0.97 g, 10.38 mol) was stirred for 2 days at room temperature under sealing. TLC showed no starting material remained. The mixture was diluted with water (6 mL) and basified with Na<sub>2</sub>CO<sub>3</sub> solid to pH 9. The aqueous solution was extracted with CH<sub>2</sub>Cl<sub>2</sub> (100 mL×5). The combined organic layers were concentrated under reduced pressure to give yellow oil. The crude product was purified by column chromatography on silica gel column, eluted with CH<sub>2</sub>Cl<sub>2</sub>/MeOH (10: 0.2) to afford a colorless solid (0.6 g, 68.1%). The solid (0.45 g, 1.13 mmol) was dissolved in dry ethanol (8 mL) under an ice-water bath, and the solution was acidified by alcoholic solution of sulfuric acid (3 mL, 0.7 mL conc. H<sub>2</sub>SO<sub>4</sub> in 28 mL ethanol) to pH 5. The mixture was stirred for 3 hours. The resulting precipitate was filtered, and the compound (10) was obtained as a white solid (0.47g, 83.2%).

<sup>1</sup>H NMR (D<sub>2</sub>O, 300 Hz):  $\delta$  8.17 (d, 1H, J=7.5 Hz), 8.12 (d, 1H), 7.76 (d, 1H), 7.53(d, 1H, ), 6.73 (br d, 1H, J= 5.4 Hz), 4.02 (brs, 1H), 3.52 (m, 1H), 3.13 (m, 1H), 1.45-1.85 (m, 4H), 1.43-1.68 (m, 3H), 1.14 (t, 3H, J= 6.8 Hz), 1.04 (t, 1H, J= 5.8 Hz). MS-EI (m/z): 170.7 (100), 171.4 (37), 340 (68), 341.2 (16), 342.2 (24). HPLC (XDB-C18, wavelength= 240 nm, CH<sub>3</sub>OH/10mmol/L CH<sub>3</sub>COONH<sub>4</sub>+0.03% TEA=62/38, 1.0 mL/min): t<sub>R</sub>=8.95 min (98.5%). Isotopic enrichment determined by MS was over 98%.

# Synthesis of $N^4$ -(7-chloroquinolin-4-yl)pentane-1, 4-diamine (11)

A solution of (5) (21.00 g, 51.4 mmol) in methanolic ammonia (210 mL) was stirred at room temperature for 8 h. The solution was evaporated to dryness to give solid. The solid was purified by column chromatography on silica gel column, eluted with  $CH_2Cl_2$ /saturated methanol ammonia (10:1) to afford (11) as a colorless solid (6.21 g, 45.8%).

#### *Synthesis of* $[{}^{2}H_{3}]$ *N*-(4-(7-chloroquinolin-4-ylamino)-pentyl)acetamide (13)

To a solution of (11) (1.2 g, 4.55 mmol) and  $Et_3N$  (0.92, 9.1 mmol) in  $CH_2Cl_2$  was added [<sup>2</sup>H<sub>3</sub>] acetyl chloride (13) (0.445 g, 5.46 mmol) slowly for 30 min. The reaction solution was stirred for 30 min at room temperature. The solution was cooled by an ice/water bath, basified by saturated NaHCO<sub>3</sub> solution and extracted with  $CH_2Cl_2$  (50 mL×5). The organic layers were combined and concentrated under reduced pressure to give a white solid. The solid was purified by column chromatography on silica gel column, eluted with  $CH_2Cl_2$ /saturated methanol ammonia (10:

0.5) to afford (13) as a white solid (1.34 g, 95.3%).

#### Synthesis of $[{}^{2}H_{5}]$ desethylchloroquine (14)

To a suspension of (13) (0.96g, 3.11 mmol) and LiAlD<sub>4</sub> (0.261 g, 6.22 mmol) in THF (9.6 mL) was refluxed for 1 h. The reaction solution was cooled to room temperature and diluted with water (50 mL). The solution was extracted with EtOAc (60 mL×4), and the combined organic layers were concentrated under reduced to give a liquid. The crude product was purified by chromatography on a silica gel column and then eluted with  $CH_2Cl_2/saturated$  methanol ammonia (10:1) to afford (14) as a white solid (0.62 g, 67.1%).

<sup>1</sup>H NMR (DMSO, 300 Hz) δ 9.11 (2H, d, J=8.4 Hz), 8.83 (1H, d, J=9.3 Hz), 7.42(1H, d, J=6.9 Hz), 8.05 (1H, d, J=1.8 Hz), 7.75 (1H, d, J= 9.0 Hz), 6.94 (1H, d, J= 7.2 Hz), 4.11(1H, brd), 2.84 (2H, m), 1.88(1H, m), 1.68 (3H, m), 1.29 (3H, d, J= 6.3 Hz). MS-EI, (m/z): 296.2 (28), 297.2 (100), 298.2 (38), 299.2 (26), 300.2 (9). HPLC (XDB-C18, wavelength= 240 nm, CH<sub>3</sub>OH/10mmol/L K<sub>2</sub>HPO<sub>4</sub>=75/25, 1.0 mL/min): t<sub>R</sub>=6.79 min (99.2%). Isotopic enrichment determined by MS was over 98%.

# *Synthesis of* $[{}^{2}H_{4}]$ *desethylhydroxychloroquine (15)*

To a stirred solution of (11) (5 g, 18.95 mmol) and  $[^{2}H_{4}]$  2-bromoethanol (1.22 g, 9.47 mmol) in methanol (100 mL) was under ice/water bath. The solution was stirred for 8 h at room temperature. TLC showed little starting material remained. The solution was co-evaporated with ethanol (150 mL×3) to give yellow solid. The solid was purified by column chromatography on silica gel column, eluted with CH<sub>2</sub>Cl<sub>2</sub>/ saturated methanol ammonia (10: 0.3) to afford (15) as a silver gray solid (1.36 g, 46.1%).

<sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 Hz)  $\delta$  8.48 (1H, d, J=5.4 Hz), 7.92 (1H, d, J=2.1 Hz), 7.70 (1H, d, J=9.0 Hz), 7.32 (1H, d, J=2.1 Hz), 7.29 (1H, d, J= 2.4 Hz), 6.38 (1H, d, J= 5.4 Hz), 5.41 (1H, brd), 3.70(1H, m), 2.74 (2H, m), 1.88(2H, m), 1.68 (2H, m), 1.29 (3H, d, J= 6.3 Hz). MS-EI, (m/z): 311.8 (100), 312.8 (43), 313.8 (16), 314.8 (6). HPLC (XDB-C18, wavelength= 240 nm, CH<sub>3</sub>OH/10mmol/L K<sub>2</sub>HPO<sub>4</sub>=65/35, 1.0 mL/min): t<sub>R</sub>=8.67 min (99.9%). Isotopic enrichment determined by MS was over 98%.

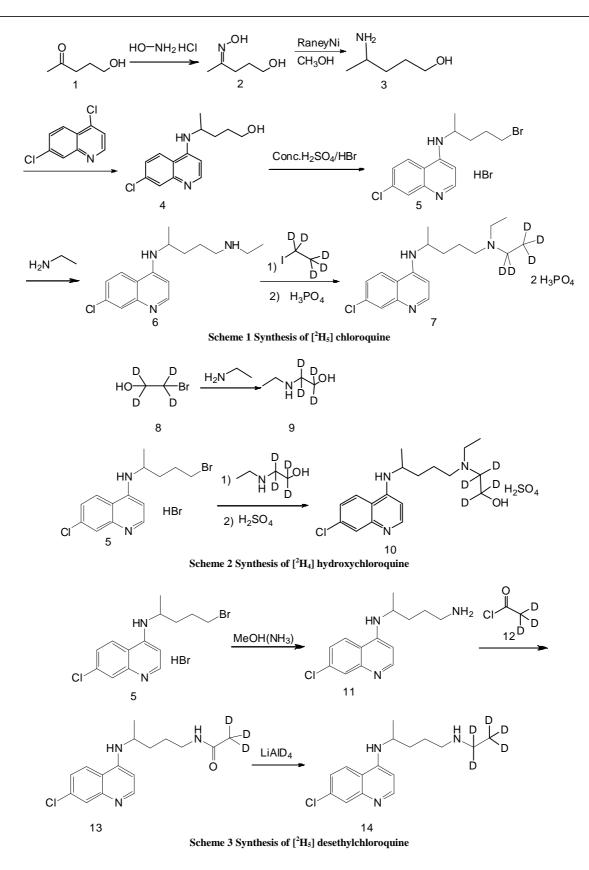
# **RESULTS AND DISCUSSION**

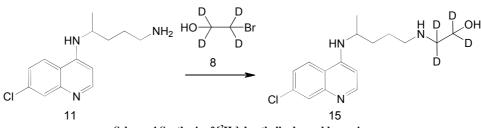
Although chloroquine, hydroxychloroquine and desethylchloroquine have been readily prepared via several synthetic routes [15-22], the synthesis of [ ${}^{2}H_{5}$ ] chloroquine, [ ${}^{2}H_{4}$ ] hydroxychloroquine, [ ${}^{2}H_{5}$ ] desethylchloroquine and their metabolites have not been described previously. Scheme 1 presents the general synthetic scheme for preparing [ ${}^{2}H_{5}$ ] chloroquine. 5-hydroxypenta-2-one (1) was condensed with hydroxylamine hydroxychloride in the presence of KOH to yield (Z)-5-hydroxypentan-2-one oxime (2). Hydrogenolysis of compound (2) with Raney nickel and hydrogen (0.8 Mpa) afforded 4-aminopentan-1-ol (3). Nucleophilic substitution of the chlorine atom in 4, 7-dichloroquinoline with 4-aminopentan-1-ol (3) at 140 °C produced 4-(7-chloroquinolin-4-ylamino)pentan-1-ol (4) [23]. Compound (4) was brominated by 48% HBr in the presence of conc. H<sub>2</sub>SO<sub>4</sub> was afford the bromoquine derivative (5), which was alkylated with 70% ethylamine solution to give diamine (6). The diamine (6) was further alkylated with [ ${}^{2}H_{5}$ ] ethyl iodide in presence of K<sub>2</sub>CO<sub>3</sub> to give [ ${}^{2}H_{5}$ ] chloroquine. After purification of the free base by column chromatography, [ ${}^{2}H_{5}$ ] chloroquine diphosphate (7) was prepared.

Scheme 2 presents the general synthetic scheme for preparing  $[^{2}H_{4}]$  hydroxychloroquine.  $[^{2}H_{4}]$  2-bromoethanol (8) was alkylated with 70% ethylamine solution to give  $[^{2}H_{4}]$  2-(ethylamino)ethanol (9), $[^{22}]$  which was treated with bromoquine derivative (5) to give  $[^{2}H_{4}]$  hydroxychloroquine. After purification of the free base by column chromatography,  $[^{2}H_{4}]$  hydroxychloroquine sulfate (10) was prepared.

Scheme 3 presents the general synthetic scheme for preparing  $[{}^{2}H_{5}]$  desethylchloroquine. The ammoniation of compound (5) in saturated methanol ammonia solution in a sealed tube afforded the diamine (11). Acylation of compound (11) was carried out with  $[{}^{2}H_{3}]$  acetyl chloride (12) in dry CH<sub>2</sub>Cl<sub>2</sub> in presence of Et<sub>3</sub>N to give  $[{}^{2}H_{3}]$  amide (13), which was reduced by LiAlD<sub>4</sub> to give  $[{}^{2}H_{5}]$  desethylchloroquine (14).

Scheme 4 presents the general synthetic scheme for preparing  $[{}^{2}H_{4}]$  desethylhydroxychloroquine (15). The diamine (11) was alkylated with  $[{}^{2}H_{4}]$  2-bromoethanol (8) to give  $[{}^{2}H_{4}]$  desethylhydroxychloroquine (15).





Scheme 4 Synthesis of [<sup>2</sup>H<sub>4</sub>] desethylhydroxychloroquine

HPLC results showed that  $[^{2}H_{5}]$  chloroquine (7),  $[^{2}H_{4}]$  hydroxychloroquine (10),  $[^{2}H_{5}]$  desethylchloroquine (14) and  $[^{2}H_{4}]$  desethylhydroxychloroquine (15) were obtained with over 98% chemical purity. Mass spectrometry analysis of compound (7), compound (10), compound (14) and compound (15) revealed over 98% deuterium enrichment.

#### CONCLUSION

The present study describes the synthesis of deuterium-labelled chloroqiune, hydroxychloroquine and their metabolites. Mass spectrometry analysis of these compounds revealed over 98% deuterium enrichment. The compounds can be used as internal standards in metabolism studie.

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