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**Research Article** 

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# Synthesis, characterization and biological evaluation of benzoxazole derivatives

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## ABSTRACT

A series of some novel benzoxazoles were synthesized and evaluated for anti microbial, anti-inflammatory and analgesic activity. The reaction of aniline compounds with ammonium thiocyanate and bromine in glacial acetic acid yield 4-thiocyanoaniline. The title compounds were synthesized by treating 4-thiocyanoaniline with o-aminophenol and carbon-di-sulphide. Their structures were confirmed by IR, <sup>1</sup>H-NMR and <sup>13</sup>C-NMR. Antimicrobial activity against bacteria and fungi ,anti-inflammatory activity and analgesic activity were studied for the synthesized compounds.

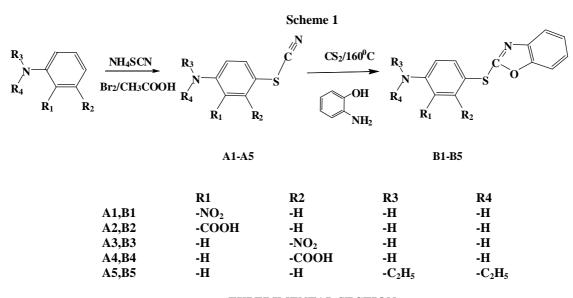
Keywords: benzoxazole, thiocyanation, anti microbial, anti-inflammatory.

#### INTRODUCTION

Recent observations suggest that substituted benzoxazoles and related heterocycles, possess potential activity with lower toxicities in the chemotherapeutic approach in man <sup>[1,2]</sup>. Careful literature survey revealed that targets containing benzoxazole moiety, either isolated from plants or accessed by total synthesis, have remarkable biological activities<sup>[3]</sup> like antimicrobial<sup>[4]</sup>, antihistaminic<sup>[5]</sup>, antiparasitics<sup>[6]</sup>, antiviral<sup>[7]</sup>, antiallergic<sup>[8]</sup>, antifungal<sup>[9,10]</sup> and antihelmintic<sup>[11]</sup> activity. Benzoxazole is used primarily in industry and research, and has no household use. Being a heterocyclic compound, benzoxazole finds use in research as a starting material for the synthesis of larger, usually bioactive structures. Benzoxazoles can be considered as structural isosteres of the naturally occurring nucleic bases adenine and guanine, which allow them to interact easily with polymers of living systems.

The title compounds were synthesized by treating thiocyano-aniline derivatives with o-aminophenol and carbondisulphide to get a new series of benzoxazole derivatives(scheme 1). The purity of the synthesized compounds were judged by their C,H and N analysis and the structures were confirmed on the basis of IR, <sup>1</sup>H-NMR, <sup>13</sup>C-NMR and mass spectral data. The anti microbial activity of the synthesized compounds was tested by using disc diffusion method. The anti-inflammatory activity of the novel compounds was determined by carrageenan induced hind paw edema using Indomethacin as a standard and the analgesic activity was evaluated by Hotplate method.

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## **EXPERIMENTAL SECTION**

All melting points were taken in open capillaries and are uncorrected. Elemental analysis was performed on a Perkin-Elmer analyzer. IR spectra<sup>[12]</sup> were recorded in KBr on Shimadzu spectrometer, <sup>1</sup>H-NMR<sup>[13]</sup> and <sup>13</sup>C-NMR<sup>[14]</sup> in DMSO-d6 on a Bruker AC-400 spectrometer using TMS as an internal standard. The microorganisms were obtained from National Chemical Laboratory, Pune.

## General procedure for the synthesis of thiocyanate (A1-A5)

The substituted/unsubstituted aniline (0.5 mol) was dissolved in acetic acid (125 ml) and the solution was added to the solution of ammonium thiocyanate (1.05 mol, 80 g) in glacial acetic acid (250 ml). This solution was cooled to  $10-20^{\circ}$  C . To this well stirred solution, a solution of bromine (25.7 ml, 0.5 mol) in acetic acid (250 ml) was added dropwise for a period of twenty to thirty minutes, and the temperature was maintained below  $20^{\circ}$ C. After the complete addition of bromine, it was kept at room temperature for ten minutes and then it was diluted with an equal amount of water. The product thus obtained was filtered, washed, dried and recrystallized from ethanol.The melting point and percentage yield are reported in table1.

#### General procedure for the synthesis of benzoxazoles(Compound B1-B5)

A mixture of thiocyanate A1-A5 (0.01 mol), o-aminophenol (0.01 mol, 1.09 g) and carbon disulphide (0.1 mol, 8 ml) was heated in an oil bath at  $160^{\circ}$  C for 6 hours. The resultant was cooled and recrystallised from ethanol.

Thiocyanate Yld (%)		M. Pt Molecular		Elemental Analysis (%) Reported (Calculated)					
		(° C)	formula	С	Н	N	0	S	
A1	76	115-116	$C_7H_5SN_3O_2$	43.18 (43.07)	2.51 (2.58)	21.29 (21.53)	16.32 (16.39)	16.39 (16.43)	195
A2	62	218-219	$C_8H_6SN_2O_2$	49.41 (49.47)	3.08 (3.11)	14.38 (14.42)	16.40 (16.48)	16.45 (16.50)	194
A3	97	139-140	$C_8H_6SN_2O_2$	49.42 (49.47)	3.05 (3.11)	14.35 (14.42)	16.39 (16.48)	15.90 (16.05)	194
A4	75	85-86	C <sub>11</sub> H <sub>14</sub> N2S	63.99 (64.04)	6.80 (6.84)	13.47 (13.58)	_	15.48 (15.54)	206
A5	80	91-92	C <sub>7</sub> H <sub>5</sub> SN <sub>3</sub> O <sub>2</sub>	43.01 (43.07)	2.40 (2.58)	21.29 (21.35)	16.42 (16.39)	16.38 (16.43)	195

Tabl	le 1	l Ana	lytical	data	a of	thiocya	nate	A1-A5
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### IR data for the compounds A1-A5

A1 (2-nitro-4-thiocyanatoaniline)  $- v_{C=N} : 2170 \text{ cm}^{-1}$ 

A2 (5-amino-2-thiocyanatobenzoicacid) - v  $_{C=N}:~2150~\text{cm}^{-1}$ 

A3 (2-amino-5-thiocyanatobenzoicacid) - v  $_{C\equiv N}:~2155~\text{cm}^{-1}$ 

A4 (N,N-diethyl-4-thiocyanatoaniline)  $v_{C=N}$ : 2210 cm<sup>-1</sup> A5 (3-nitro-4-thiocyanatoaniline)  $v_{C=N}$ : 2257 cm<sup>-1</sup>

Benzoxazole	Yld (%)	M. Pt Molecular		Elemental Analysis (%) Reported (Calculated)					M wt
		(° C)	formula	С	Н	N	0	S	
B1	73	162-163	C13H9SN3O3	54.32	3.10	14.54	16.61	11.28	287
DI	15	102-105	C13H95IN3O3	(54.35)	(3.16)	(14.63)	(16.71)	(11.16)	207
B2	86	186-187	$C_{14}H_{10}SN_2O_3$	58.90	3.48	9.70	16.72	11.18	286
<b>D</b> 2	80 180-187	100-107	$C_{14}\Pi_{10}SIN_2O_3$	(58.73)	(3.52)	(9.78)	(16.76)	(11.20)	280
B3	71	115-116	$C_{14}H_{10}SN_2O_3$	58.78	3.56	9.69	16.69	11.26	286
Б3	/1	113-110	$C_{14}\Pi_{10}SN_2O_3$	(58.73)	(3.52)	(9.78)	(16.76)	(11.20)	200
B4	61	101-102	CUNSO	68.32	6.15	9.30	5.28	10.76	298
D4	01	101-102	$C_{17}H_{18}N_2SO$	(68.42)	(6.08)	(9.39)	(5.36)	(10.75)	298
В5	79	122-123	CUSNO	54.31	3.11	14.51	16.65	11.15	287
ы	19	122-123	$C_{13}H_9SN_3O_3$	(54.35)	(3.16)	(14.63)	(16.71)	(11.16)	207

#### Table 2 Analytical data of benzoxazole (B1-B5)

**Compound B1(4-(benzo[d]oxazol-2-ylthio)-2-nitroaniline):**IR(KBr) cm<sup>-1</sup> : 3306(NH<sub>2</sub>), 1639(C=N str), 3189(NH), 3086(aromatic) .<sup>1</sup>H-NMR :  $\delta 6.8 - 7.5$  (Ar-H, multiplet), $\delta 2.88$  (Ar-NH<sub>2</sub>, singlet). <sup>13</sup>C-NMR :  $\delta 110-180$  (Ar-C),  $\delta 148.6$  (C=N).

**Compound B2(5-amino-2-(benzo[d]oxazol-2-ylthio)benzoic acid):** IR(KBr) cm<sup>-1</sup> : 3413(NH<sub>2</sub>), 1629(C=Nstr), 3207(NH), 2565(OH str). <sup>1</sup>H-NMR :  $\delta$  6.4 – 7.4 (Ar-H, multiplet),  $\delta$  9.9 (-COOH, singlet). <sup>13</sup>C-NMR :  $\delta$  108-149 (Ar-C),  $\delta$ 157 (C=N).

**Compound B3(2-amino-5-(benzo[d]oxazol-2-ylthio)benzoic acid):** IR(KBr) cm<sup>-1</sup> : 3425(NH<sub>2</sub>), 1601(C=N str), 2577(OH str), 738(C=C bending).<sup>1</sup>H-NMR :  $\delta$  6.0 – 9.0 (Ar-H, multiplet), $\delta$  3-5.5 (Ar-NH<sub>2</sub>, singlet).<sup>13</sup>C-NMR :  $\delta$  122.7 (Ar-C),  $\delta$  147 (C=N).

**Compound B4(4-(benzo[d]oxazol-2-ylthio)**-*N*,*N*-diethylaniline):IR(KBr) cm<sup>-1</sup> : 3439(NH<sub>2</sub>), 1583( C=N str), 3013(aromatic) .<sup>1</sup>H-NMR :  $\delta$  6.9 – 7.9 (Ar-H, multiplet),  $\delta$  1.2 (C<sub>2</sub>H<sub>5</sub>, singlet). <sup>13</sup>C-NMR :  $\delta$  110-124 (Ar-C) .

**Compound B5(4-(benzo[d]oxazol-2-ylthio)-3-nitroaniline):**IR(KBr) cm<sup>-1</sup> : 3424(NH<sub>2</sub>), 1418(NO<sub>2</sub>), 1623( C=N str), 2937(aromatic) .<sup>1</sup>H NMR :  $\delta$ 7.0 – 7.9 (Ar-H, multiplet), $\delta$ 2.2 (Ar-NH<sub>2</sub>, singlet),  $\delta$ 11.3 (NH, singlet). <sup>13</sup>C-NMR :  $\delta$  109-153 (Ar-C) ,  $\delta$  152(C=N).

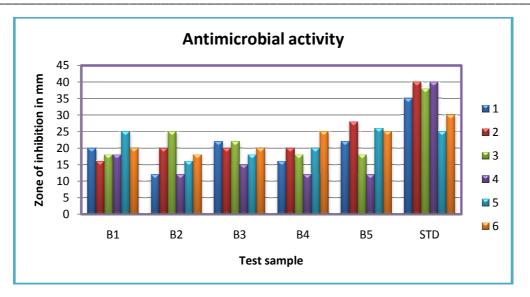
#### **RESULTS AND DISCUSSION**

#### **Anti-microbial Activity**

The anti-microbial activity for the sample was carried out by Disc Diffusion Technique<sup>[15]</sup>. The test microorganisms(*Staphylococcus aureus, Bacillus subtilis, Escherichia coli, Pseudomonas aeruginosa, Candida albicans, Aspergillus niger*) maintained by periodical subculturing on nutrient agar and sabouraud dextrose agar medium for bacteria and fungi respectively. The effects produced by the sample were compared with the effect produced by the positive control (Reference standard ciprofloxacin 5  $\mu$ g/disc for bacteria; Nystatin 100 units/disc for fungi).

S.No	Name of the microorganisms		Zone	of Inhi	bition	in mm	ı
	Name of the microorganisms		B2	B3	B4	B5	Std
1.	Staphylococcus aureus	20	12	22	16	22	35
2.	Bacillus Subtilis	16	20	20	20	28	40
3.	Escherichia Coli	18	25	22	18	18	38
4.	Pseudomonas aeruginosa	18	12	15	12	12	40
5.	Candida Albicans	25	16	18	20	26	25
6.	Aspergillus Niger	20	18	20	25	25	30

Table 2	Antimicrobial	activities	of the	synthesized	compounds
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 <sup>1.</sup>Staphylococcus aureus
 2.Bacillus Subtilis
 3.Escherichia Coli
 4.Pseudomonas aeruginosa
 5.Candida Albicans
 6.Aspergillus Niger

#### **Anti-inflammatory Activity**

#### Carrageenan induced hind paw edema:

Albino rats of either sex weighing 150-200 gms were divided into six groups of six animals each. The dosage of the drugs administrated to the different groups were as follows: Group 1 – Control received normal saline, Group 2 to 16 received test in a dose of 50 mg/kg and Group 17-Indomethacin(10mg/Kg). All the drugs were administrated orally.

After one hour of the administration of the drugs, dose 0.1 ml of 1% w/v carrageenan solution in normal saline was injected into the subplantar tissue of the left hind paw of the rat and the right hind paw served as the control. The paw volume of the rats were measured in the digital plethysmograph(Ugo basile, Italy) at the end of 0, 60, 120 and 180 min. The increase in paw edema of the treated group was compared with that of the control and the inhibitory effect of the drugs were studied. The relative potency of the drugs under investigations were calculated based upon the percentage inhibition of the inflammation.

<b></b>	Control(increase in paw volume in 3 <sup>rd</sup> hour)	Test(increase in paw volume in 3 <sup>rd</sup> hour)	X 100
Percentage Inhibition =			A 100

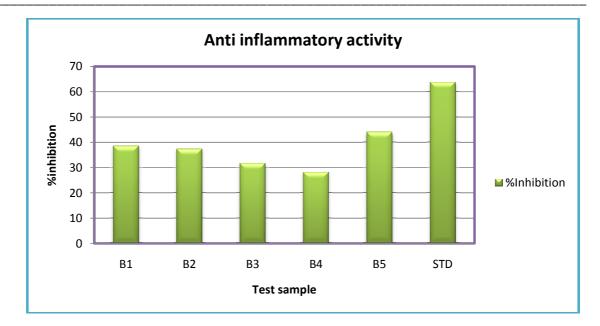
Control (increase in paw volume in 3<sup>rd</sup> hour).

Table 3 Anti inflammatory activity of the synthesized compounds

Treatment	Dose M g/kg p.o.	Paw volume increase after 3 hr(ml)	Percentage inhibition
control	5 ml/kg	$111.61 \pm 10.56$	-
B1	50	$68.62 \pm 5.32$	38.51
B2	50	$66.94 \pm 7.26$	37.35
B3	50	$76.24 \pm 6.42$	31.69
B4	50	$80.18\pm5.68$	28.16
B5	50	$62.44 \pm 5.22$	44.05
Indomethacin	10	$40.4 \pm 3.62$	63.80

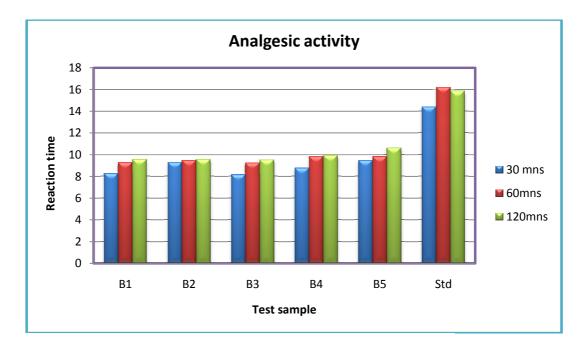
\*P<0.001 values are expressed as ± SEM. Number of animals using are 6 in each group.

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## Analgesic activity

The analgesic activity of the given sample was evaluated by using Hotplate method. The albino mice of either sex were used, the animals were divided into nine groups of 5 animals each. Group 1 received normal saline(1ml/kg), group 2 received standard (pentazocine 10 mg/kg) intraperitonealy, groups 3 to 9 received the given extract (50 mg/kg) orally. Before administrating the drug, basal reaction time was studied by placing the animals in hotplate and the parameters such as paw licking, jumping response were noted. The maximum cutoff time is 15 sec. After half an hour of administration of the drug, the reaction time was noted and compared.



			Dose	Reaction time(in sec)				
S.No	Groups	Dava		Before	ter administra	tion		
5.110	Groups	Drug		(mg/ administratn of drug				
			kg)	of drug	30 mins	60 mins	120 mins	
1.	Control	Saline	1ml/kg	4.41±0.16	4.42±0.20	4.48±0.20	4.43±0.17	
2.	Test-1	B1	50	4.38±0.14	8.26±0.18	9.28±0.26	9.56±0.32	
3.	Test-2	B2	50	4.28±0.22	9.28±0.18	9.46±0.26	9.56±0.32	
4.	Test-3	B3	50	4.12±0.24	8.16±0.14	9.24±0.24	9.52±0.18	
5.	Test-4	B4	50	4.42±0.34	8.76±0.18	9.84±0.26	9.92±0.32	
6.	Test-5	B5	50	4.44±0.24	9.46±0.42	9.82±0.32	10.62±0.24	
7.	standard	pentazocine	10	5.42±0.16	14.4±0.32	16.2±0.18	15.86±0.28	

Table 4 Analgesic activity of the synthesized compounds

Mean  $\pm$  S.E.M, n=5.

#### DISCUSSION

The thiocyanates A1-A5 were synthesized in good yield by the reaction of aniline derivatives with ammonium thiocyanate and  $Br_2/CH_3COOH$  under ice-cold condition. Compounds A1-A5 on reaction with o-aminophenol in the presence of carbondisulphide afforded compounds B1-B5. The purity and homogeneity of all the synthesized compounds were confirmed by their sharp melting points (uncorrected) and column chromatography. The chemical structures were confirmed by IR, <sup>1</sup>H-NMR and <sup>13</sup>C\_NMR techniques. The aromatic stretching frequencies for all the derivatives were found to be at the range of 2900-3100 cm<sup>-1</sup>. The presence of NH stretching was confirmed by the peaks at 3100-3200 cm<sup>-1</sup>. Also <sup>1</sup>H-NMR spectra were useful for identifying protons. The peaks at the frequency range 6.0 – 9.0 confirms the aromatic protons and 2.2-5.8 confirms the NH<sub>2</sub> protons. From the microbiological data, it was observed that compounds. The anti-inflammatory activity study showed that compound B5 has significant effect over carrageenan induced hind paw edema. On percentage protection basis, compound B2 showed 44.05%, while Indomethacin showed 63.80% when compared to control. Compound B5 proved to possess potential analgesic activity.

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