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Review Article

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Structure Activity Relationship and Mechanism of Action of Hydrazide Derivatives as Antimicrobial Molecule

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ABSTRACT

Hydrazide and their heterocyclic derivatives exhibiting wide range of biological activities. Among different biological activities antimicrobial, antitubercular, antiamoebic and antimalarial are noteworthy. Further, presence of different substituents and derivatization to heterocyclic ring of hydrazide derivatives affect biological outcomes to a great extent. Therefore, researchers have synthesized compounds containing hydrazides and their heterocyclic derivatives with variable substituents as target molecules and evaluated their biological activities. Present review describes the antimicrobial importance of hydrazide derivatives along with their structure activity relationship. Moreover, we have put a light on targets of antimicrobial hydrazide derivatives through which they induce biological effects. Present review will be helpful in designing and structure activity relationship study of bioactive hydrazide derivatives.

Keywords: Antimicrobial, Antitubercular, Antiamoebic, Antimalarial, Hydrazides, Heterocyclics.

INTRODUCTION

Hydrazide-Hydrazone derivatives containing –NHN=CH- moiety represent an over whelming and rapid developing field in modern medicinal chemistry. [1] Reported data indicated that hydrazide derivatives have significant biological activities such as anti-inflammatory [2], antimalarial [3], antimicrobial [4], antileishmanial [5], anticonvulsant [6], antitubercular [7] and antitumor [8]. Examples of clinically used drugs having hydrazide moiety presented in Table 1.

Further reported data and studies performed by us revealed that antimicrobial activity is most significant among different activities for hydrazide derivatives [10-13]. Therefore, in present study we attempted to compile reported data of most active hydrazides having antimicrobial activity and their mechanism of action. Present review will help the researchers in understanding the structure activity relationship (SAR) and mechanism of action of hydrazide derivatives.

Mechanism of action of existing antimicrobial hydrazides and related drugs

Isoniazide (hydrazide of nicotinic acid) activation leads to inhibition of the synthesis of mycolic acid, an important component of the mycobacterial cell wall. Two enzymes involved in the cycle of the fatty acid biosynthesis, namely an enoyl-acyl carrier protein reductase (InhA) and β -ketoacyl- acyl carrier protein synthase, are believed to be targets of the activated isoniazid. [14]

Thiazolidinones, hydrazide derivative inhibit the biosynthesis of the peptidoglycan polymer essential for cell wall *via* inhibition of MurB enzyme of bacteria. MurB enzyme causes reduction of enolpyruvyl uridine diphosphate *N*-acetylglucosamine (EP-UNAG) to uridine diphosphate *N*-acetylmuramic acid (UNAM), is a unique target for antibacterial activity of thiazolidinones. [15]

Hydrazide derivatives viz. 2-azetidinones (β-lactam antibiotics) are one of the most prescribed chemotherapeutic agents used in medicine to treat bacterial infections. 2-Azetidinone molecules act by forming a covalent adduct with membrane-bound bacterial transpeptidase, which are also known as penicillin-binding proteins, involved in the biosynthesis of cell walls. On the basis of this mechanism, inhibitors prevent the construction of cell wall and eventually lead to cell lysis and death. Moreover, due to their β-lactamase inhihibitory action, 2-azetidinones based heterocycles represent an attractive target against microorganisms. [16]

S. No.	Name of Drug	Biological activity	Chemical Structures
1.	Isoniazide	Antituberculosis	O NH H ₂ N
2.	Nifuroxazide	Antibiotic	HO-N-N H N-N H ONO ₂
3.	Nifurtimox	Antiameobic	
4.	Isocarboxazide	Antidepressant	$\begin{array}{c} O \\ CH_2 \cdot NH - NH - \overset{\circ}{C} \end{array}$
5.	Iproniazide	Antituberculosis	CH ₃ C-NH-NH—CH CH ₃ CH ₃

Table 1 Examples of drugs containing hydrazide moiety [9]

Hydrazide derivative as antibacterial and antifungal agents

Biphenyl-4-carboxylic acid hydrazide-hydrazones derivatives have been synthesized by Deep *et al.* All the compounds were tested for their *in vitro* antimicrobial activity against two Gram-negative strains (*Escherichia coli* and *Pseudomonas aeruginosa*) and two Gram-positive strains (*Bacillus subtilis* and *Staphylococcus aureus*) and fungal strain *Candida albicans* and *Aspergillus niger* using tube dilution method, all synthesized compounds exhibited promising activity (MIC = $25 \mu M/mL$). Result of antimicrobial activity indicated that presence of *p*-NO₂ and *p*-Cl increased antifungal and antimicrobial activities. [17]

1,
$$R = CI \text{ or } NO_2$$

Wang *et al.* synthesized vanillic acyl hydrazone derivatives and evaluated their *in vitro* antibacterial activities against *E. coli*, *P. aeruginosa*, *S. aureus*, and *B. subtilis* by broth dilution method. The result of antimicrobial evaluation indicated that compound (*E*)-*N*-(3,5-dichloro-2-hydroxybenzylidene)-4-hydroxy-3-methoxy benzohydrazide (**2**) was most active against *E. coli* among synthesized derivatives (MIC = 0.39 μ g/ml). The highest activity of compound **2** may be due to the formation of hydrogen bond with target site by hydroxyl group present in synthesized derivatives. [18]

Altintop *et al.* synthesized hydrazone derivatives by condensation of 2-[(1-methyl-1H-tetrazol-5-yl)thio)] acetohydrazide with aromatic aldehydes/ketones and evaluated their *in vitro* anticandidal activity against *C. albicans, C. krusei, C. parapsilosis by* micro dilution method. The result of antimicrobial evaluation showed that compound N'-(4-(dimethyl amino)benzylidene)-2-((1-methyl-1H-tetrazol-5-yl)thio)acetohydrazide (3) was most active among synthesized derivatives (MIC = 0.05 mg/ml) and comparable with standard drug Ketoconazole (MIC= 0.031 mg/ml) against *C. albicans*. [19]

Pieczonka *et al.* reported synthesis of 3-oxido-1*H*-imidazole-4-carbohydrazides and evaluated their *in vitro* antibacterial activities against *E. coli*, *S. aureus* and *S. epidermidis* by micro dilution method. Compound 1-benzyl-5-methyl-N'-[(Z)-(5-nitro-2-furyl) methylidene]-1*H*-imidazole-4-carbohydrazide-3-oxide (4) was found to be most effective against *S. epidermidis* having MIC value of 4 μ g/ml. Antimicrobial results analysis showed that presence of furyl ring with electron withdrawing nitro group is responsible for high potency of compound 4. [20]

Kumar *et al.* synthesized hydrazone derivatives and evaluated their *in vitro* antibacterial activity against *S. aureus*. The most active compound was found to be (*E*)-5-bromo-2-hydroxy-N'-(1-(1-methyl)-1*H*-indole-2-yl)ethylidene)benzohydrazide (**5**) (MIC = 1 μ g/ml)³⁴ among the synthesized derivatives due to presence of electron donating hydroxyl and electron withdrawing bromo groups. [21]

Plech *et al.* synthesized a novel series of 1,2,4-triazole-ciprofloxacin hybrids and evaluated their antibacterial activities against *M. luteus, E. coli, P. mirabilis, P. aeruginosa, S. aureus, S. epidermidis, B. subtilis* and *B. cereus* by agar diffusion method . Compound 1-cyclopropyl-7-(4-((4-(2,4-dichlorophenyl)-3-(3-hydroxyphenyl)-5-thioxo-4,5-dihydro-1H-1,2,4-triazol-1-yl)methyl)piperazin-1-yl)-6-fluoro-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (6) was found to be most potent (MIC = 0.35 μ M/ml) against *S. aureus* among the synthesized derivatives due to the presence of *m*-hydroxyl phenyl moiety attached with triazole ring. [22]

Emary *et al.* reported a novel series of indeno[2,1-e]pyrazolo[3,4-b]pyrazin-5-one derivatives and studied their antifungal activities against *Trichophyton rubrum*, *Aspergillus flavus* by broth dilution method. Compound (Z) 2-(3-methyl-1-phenylindeno[2,1-e]pyrazolo[3,4-b]pyrazin-5(1H)-ylidene) hydrazine carbothioamide (T) was found to be most active (MIC = 0.3 mg/ml) among synthesized derivatives whereas, MIC value of standard drug Clotrimazole was 0.08 mg/ml against T. Tubrum. [23]

Manjunatha *et al.* synthesized a series of 1,3,4-oxadiazole derivatives and evaluated their antibacterial activities against *S. aureus*, *E. coli*, *P. aeruginosa* and *K. pneumonia* by serial plate dilution method. Compound 5-(4-isobutylphenyl)ethyl-3-(morpholin-4-ylmethyl)-1,3,4-oxadiazole-2(3*H*)-thione (8) was found to be most potent having MIC value 6.25 μ g/ml against tested strains. Structure activity relationship (SAR) of synthesized derivatives revealed the importance of hydrophobic 4-isobutylphenyl ethyl moiety in improving the antibacterial activity. [24]

Patel *et al.* reported synthesis of thiazolidinone derivatives and evaluated their antimicrobial activities against different bacteria *i.e. S. aureus, B. cereus, E. coli, P. aeruginosa, K. pneumoniae, S. typhi, P. vulgaris, S. flexneri* by micro dilution method. The results of antimicrobial activity showed that compound 4-(4-{4-[2-(2,4-dichlorophenyl)-5-(4-ethylpiperazin-1-ylmethyl)-4-oxothiazolidin-3-yl]-phenyl amino}-6-dimethylamino-[1,3,5] triazin-2-yloxy)-1-methyl-1H-quinolin-2-one (9) was most potent against *S. aureus* among all synthesized derivatives (MIC = 6.25 μ g/ml). SAR analysis of synthesized derivatives indicated that high potency of compound 9 was due to presence of dichlorophenyl moiety. [25]

Seelam *et al.* synthesized a novel series of fused 1,2,4-triazole derivatives and evaluated their antibacterial activities against *B. subtilis, B. thuringiensis, E. coli, P. aeruginosa* by agar diffusion method. Compound 3-(4-chlorophenyl)-6-(p-tolyl)-3,3a-dihydo-2*H*-pyrazolo[3',4':4,5] thiazolo[3,2-b] [1,2,4]-triazole (10) was found to have better activity against all the tested microorganism than other synthesized derivatives (MIC =3.125 μ g/ml). Analysis of activity data showed that high potency of compound 10 due to the presence of electron-negative chlorine atom. [26]

Gaikwad *et al.* reported thiazole substituted benzotriazole derivatives and evaluated their antibacterial activity against *S. aureus*, *E. coli*, *B. subtilis* and *P. aeruginosa* by micro broth dilution method. The results of antimicrobial activities showed that compound 1-(1-(3,5-bis(trifluoromethyl)phenyl)-2-(1*H*-1,2,4-triazol-1-yl)ethylidene)-2-(4-(4-fluorophenyl) thiazol-2-yl)hydrazine (11) was most active one against *P. aeruginosa*, among all the synthesized derivatives (MIC = $16 \mu g/ml$). The high potency of synthesized compound (11) due to presence of electron withdrawing trifluoromethyl group at *meta* position of phenyl ring. [27]

Gaikwad *et al.* also reported antifungal activity of synthesized derivatives against fungal strains *A. niger*, *C. albicans*. The results of antifungal activities showed that compound 1-(1-(3,5-bis (trifluoro methyl) phenyl)-2-(1H-1,2,4-triazol-1-yl)ethylidene)-2-(4-(4-chlorophenyl)thiazol-2-yl)hydrazine (12) was most potent against *A. niger* among all synthesized derivatives (MIC = 16 μ g/ml). The present results once again indicated the presence of electron-negative trifluoromethyl group in improving antifungal activity of synthesized hydrazide derivatives. Moreover, above study revealed the significance of p-fluorophenyl and p-chlorophenyl ring, attached at 4^{th} position of thiazole ring in antibacterial and antifungal activities, respectively. [27]

Ozkay *et al.* reported benzimidazole derivatives bearing hydrazone moiety and evaluated their antibacterial activity against *P. vulgaris*, *S. typhimurium*, *K. pneumoniae* and *P. aeruginosa* by broth micro dilution method. The results of antibacterial activity showed that compound (*E*)-4-(1*H*-benzo[d]imidazol-2-yl)-N'-(4-chloro benzylidene) benzohydrazide (13) was found to be most potent against *S. typhimurium* among all synthesized derivatives (MIC = 6.25 µg/ml). High potency of aforementioned compound was due to presence of electron withdrawing chlorine group. [28]

$$\begin{array}{c|c}
N & O \\
N & N \\
13
\end{array}$$

Refat *et al.* synthesized (benzochromen-2-yl)ethylidene) acetohydrazide derivatives and evaluated their *in vitro* antibacterial activities against *S. aureus* and *E. coli* by broth micro dilution method. Compound 3-oxo-N'-(1-(3-oxo-3H-benzo[f]chromen-2-yl)ethylidene)-3H-benzo[f]chromene-2-carbohydrazide (14) was found to be most active one against *S. aureus* with MIC value = 62.5 μ g/ml among the synthesized derivatives. The antimicrobial results signify the presence of pyran ring in improving the antibacterial activity of synthesized derivatives. [29]

A series of hydrazide derivatives were evaluated for their *in vitro* antibacterial activities against *E. coli*, and *S. aureus*. Compound N-((2R,3R)-3-chloro-2-(4-hydroxyphenyl)-4-oxoazetidin-1-yl)-3-(4-chlorophenyl)-1H-pyrazole-5-carboxamide (15) was found to be most against *E. coli* having MIC value of 0.010 μ g/ml. High potency of compound (15) may be due to the presence of hydrogen bond forming hydroxy and electron negative chloro groups. Result analysis of present study showed the significance of azetidinone and pyrazole rings in antibacterial and antifungal activities, respectively. [30]

Pathak *et al.* synthesized 3-(4-chlorophenyl)- 4-substituted pyrazoles derivatives and evaluated their *in vitro* antifungal activities against *C. krusei*, *C. neoformans*, *A. niger*, *A. flavus* by national committee for clinical laboratory standards. Compound 3-(4-chlorophenyl)-*N*-(2-(2-hydroxyphenyl)-4-oxothiazolidin-3-yl)-1*H*-pyrazole-5-carboxamide (16) was found to be most potent having MIC value of 6.3 μg/ml. Antimicrobial result analysis showed that hydrogen bond forming hydroxyl group and electron withdrawing chlorine group are responsible for high potency of synthesized derivatives. [30]

Lee *et al.* synthesized hydrazide derivatives as selective inhibitors of *S. aureus* β -ketoacyl acyl carrier protein synthase III and evaluated their *in vitro* activity by serial broth dilution method. (*E*)-2, 4-dichloro-*N*'-(2, 3, 4-trihydroxybenzylidene)benzohydrazide (17) was found to be most potent (MIC = 2 μ g/ml) among the synthesized derivatives. The highest activity of compound (17) may be due to formation of hydrogen bond by hydroxyl groups with target site. [31]

Kumar *et al.* synthesized benzylidene hydrazides analogues and evaluated their *in vitro* antifungal activities against *A. niger* and *C. albican* by tube dilution method. The result of antifungal evaluation indicated that *N'*-benzylidene-3, 5-dimethoxybenzohydrazide (**18**) was most effective against *A. niger* among all synthesized derivatives (MIC = 1.60).

 μ M/ml). The potency of synthesized derivatives (18) may be due to presence of electron donating methoxy substituent on the phenyl ring. [32]

Kumar *et al.* synthesized of benzylidene hydrazide derivatives and evaluated their *in vitro* antibacterial activities against *S. aureus*, *B. subtilis*, and *E. coli* by tube dilution method. Compound *N*'-benzylidenedodecanehydrazide (19) was found to be most active against *S. aureus* among synthesized derivatives (MIC = $1.49 \mu M/ml$). The present results depicted the importance of long chain fatty acid in improving antibacterial activity of synthesized hydrazide derivatives. [32]

$$H_3C(H_2C)_9H_2C$$
 $NHN=CH$
 H
 H
 H
 H

Malhotra *et al.* reported a novel series of (Z)-2-(5-(biphenyl-4-yl)-3-(1-(imino)ethyl)-2,3-dihydro-1,3,4-oxadiazol-2-yl)phenol derivatives and evaluated their *in vitro* antibacterial activities against S. *aureus*, E. *coli*, B. *subtilis* and P. *aeruginosa* antifungal activity against C. *albican* and A. *niger* by disc-diffusion method. The results of antibacterial activity showed that compound (Z)-2-(5-(biphenyl-4-yl)-3-(1-(4-nitrophenylimino)ethyl)-2,3-dihydro-1,3,4-oxadiazol-2-yl)phenol (Z) was found to be most potent against Z0 and Z1 may be due to the presence of electron withdrawing nitro group at Z2 position of phenyl ring. [33]

$$\begin{array}{c|c} OH & \\ \hline \\ N-N & O \\ \hline \\ N & O^{-1} \\ \hline \\ 20 & O^{-1} \\ \end{array}$$

Tajudeen *et al.* reported isonicotinohydrazide and evaluated their antibacterial activities against *B. subtilis*, *S. aureus*, *S. pyogenes*, *E. faecalis*, *E. coli*, *K. pneumonia* by agar diffusion and macro dilution tube method. Compound (*E*)-*N*'-(3,5-dimethoxybenzylidene)isonicotinohydrazide (21) was found to be potent (MIC = 250 μg/ml) against all microorganisms as compared to other synthesized derivatives, due to the presence of electron donating OCH₃ group at *meta* position of phenyl ring. [34]

Thomas and their research group reported 4-hydroxy-8-trifluoromethyl-quinoline derivatives and evaluated their antibacterial activities against *E. coli*, *S. aureus*, *P. aeruginosa*, *K. pneumoniae* and *S. pyogenes* using serial plate dilution method. Compound 2-{[4-hydroxy-8-(trifluoromethyl) quinolin-3-yl] carbonyl}-N-(4-methoxyphenyl)hydrazine carboxamide (22) showed better antibacterial activity against *K. pneumoniae* among all synthesized derivatives (MIC = 0.1 μ g/ml). The highest potency of compound (22) was due to the presence of *p*-methoxy substituted phenyl and trifluoromethyl group at quinoline ring. [35]

Zhang *et al.* synthesized benzimidazole derivatives and evaluated their antimicrobial activities against different bacteria *i.e.* Methicillin-resistant *S. aureus*, *B. subtilis*, *M. luteus*, *E. coli*, *S. dysenteriae*, *P. aeruginosa*, *E. typhosa* by micro broth dilution method. N^2 , N^5 -bis((1-decyl-1H-benzoimidazol-2-yl)methyl)-3,4-dimethylthiophene-2,5-dicarbohydrazide (23) was found to be most potent against *S. aureus* among all synthesized derivatives (MIC = 2 µg/ml). The SAR of above antimicrobial results indicated the importance of long chain alkyl group in improving antibacterial activity. Increased potency of compound 23 is due to the high lipophilicity and hence they can easily penetrate inside lipophilic cell wall. [36]

Kocyigit-Kaymakcioglu *et al.* synthesized 3-acetyl-2,5-disubstituted-2,3-dihydro-1,3,4 oxadiazoles derivatives and evaluated their *in vitro* antifungal activity against *C. albicans* using the micro-dilution broth assay. (*E*)-1-(2-(4-(dimethylamino)styryl)-5-(4-fluorophenyl)-1,3,4-oxadiazol-3(2*H*)-yl)ethanone (**24**) was found to be most effective having MIC value 125 μ g/ml and equal to standard drug Ketoconazole (125 μ g/ml) against *C. albicans*. [37]

Narang *et al.* reported (naphthalen-1-yloxy)-acetic acid benzylidene/(1-phenyl ethylidene)hydrazide derivatives and evaluated their *in vitro* antibacterial activities *against S. aureus, B. subtilis,* and *E. coli* by serial tube dilution method. Compound (E)-N'-(4-hydroxy-3-methoxybenzylidene)-2-(naphthalen-1-yloxy)acetohydrazide (25) was found to be most active one against *B. subtilis* among all synthesized derivatives (MIC = 1.45 μ g/ml). SAR studies of synthesized hydrazide derivatives indicated the importance of hydrogen bond forming hydroxyl group and electron donating methoxy group in improving antibacterial activity of synthesized derivatives. [38]

In same study synthesized hydrazide derivatives were also evaluated their *in vitro* antifungal activities against *C. albicans* and *A. niger*. The results of antifungal activity showed that compound *N*'-(1-(2-bromo-4-cyanophenyl)ethylidene)-2-(naphthalen-1-yloxy)acetohydra -zide (26) was most effective against *C. albicans*

among all synthesized derivatives (MIC = $2.13 \mu g/ml$) due to presence of electronegative o-Br and p-CN substituent

among all synthesized derivatives (MIC =
$$2.13 \mu g/ml$$
) due to presence of electronegative *o*-Br and *p*-CN substituent on phenyl ring. [38]

Tasneem et al. reported synthesis of schiff bases and azetidinones derivatives with 1,2,4-triazoles and screened their antibacterial activity against B. subtilis, S. aureus, E. coli, P. aeruginosa, E. faecalis, K. pneumoniae. The results of antibacterial activity showed that compound 2-[4-(1-acetyl-5-phenyl-4,5-dihydro-1*H*-pyrazol-3-yl)-phenyl]-4-[3chloro-2-(4-ani-syl)-4-oxo-azetidin-1-yl]-5-methyl-3-oxo-2,4-dihydro-[1,2,4]-triazole (27) was most effective against all microorganisms among all synthesized derivatives (MIC = 12 µg/ml). The present antimicrobial result revealed the importance of 2-azetidinone and p-methoxy phenyl ring in contributing antibacterial activity of synthesized derivatives. [39]

ZA Kaplancikli synthesized oxadiazole derivatives and evaluated their in vitro antifungal activities against C. albicans, C. glabrata, C. tropicalis, C. krusei, C. parapsilosis, and C. albicans by micro broth dilution method. Compound 2-[5-[(pyrimidin-2-ylthio)methyl]-1,3,4-oxadiazol-2-ylthio]acetophenone (28) was most active (MIC = 0.007 mg/ml) among synthesized derivatives, whereas MIC value of standard drug Ketoconazole was 0.001 mg/ml against C. glabrata. [40].

Khidre et al. synthesized 1- substituted amino-4,6-dimethyl-2-oxo-pyridine-3-carbonitrile analogues and evaluated their antibacterial and antifungal activities against S. aureus, B. subtilis, P. aeruginosa, E. coli, A. fumigatus, G. candidum, C. albicans and S. racemosum by agar well diffusion method. Compound 29(a-b) showed better antibacterial and antifungal activities as compared to other synthesized derivatives against S. aureus (MIC = 24.4 and 26.4 µg/ml) and A. fumigates (MIC = 19.2 and 22.5 µg/ml). SAR study showed the significance of bromine substituted phthalimide ring in improving antimicrobial activity [41].

29a, R = H and 29b, R = Br

Secci et al. reported (4-(4-iodophenyl)-thiazol-2-yl) hydrazine derivatives and evaluated their antifungal activities against C. albicans and C. krusei by Microtiter plate method. (E)-2-(2-(1-(furan-2-yl)ethylidene)hydrazinyl)-4-(4-

iodophenyl)thiazole (30) was found to be most potent among all synthesized derivatives (MIC value = 0.25-2.00 μ g/ml) and even more active as compare to standard drug clotrimazole (MIC= 2 μ g/ml) against *C. albicans*. SAR study of synthesized derivatives showed incorporation of furan ring improved antifungal activity [42].

Gupta *et al.* reported salicylaldehyde schiff bases of 2-aminopyridine derivatives and evaluated their antibacterial activity against *S. aureus* and *E. coli* by serial dilution method. (E)-4-bromo-2-((pyridin-2-ylimino)methyl)phenol (31) was found to be most potent against *S. aureus* among all synthesized derivatives (MIC value = $0.625 \,\mu \text{g/ml}$). Antimicrobial results once again proved the importance of electron negative bromine group and hydrogen bond forming hydroxyl group (with target sites) in enhancing antibacterial activity [43].

Malladi *et al.* synthesized schiff bases analogues and screened their antibacterial activities against *S. aureus*, *B. subtilis*, *E. coli* and *P. aeruginosa* by serial dilution method. Compound 5-ethyl-4-(((3-(4-fluorophenyl)-1H-pyrazol-4-yl)methylene)amino)-4H-1,2,4-triazole-3-thiol (32) showed better antibacterial activity against *S. aureus* (MIC =3.125 μ g/ml). High potency of compound (32) due to the presence of electron withdrawing fluoro group on phenyl ring [44].

Hydrazide as antimalarial agents

Thuy *et al.* synthesized quinolinyl hydrazone derivatives and screened their *in vitro* antimalarial activities to chloroquine-sensitive (T96) and chloroquine-resistant (K1) strains of *P. falciparum*. Among the synthesized compounds, 4-((E)-((3-((E)-1-(2-(7-chloroquinolin-4-yl)hydrazono)ethyl)phenyl)imino)methylphenol (33) exhibited strong antimalarial activity at IC₅₀ of 103.4 ng/mL and 18.76 ng/mL against both strains of *P. falciparum*. The high potency of compound (33) may be due to presence of hydroxyl group, as it is normally involved in H-bond formation with target [45].

Acharya *et al.* synthesized 1,3,5-trisubstituted pyrazolines derivatives and evaluated their *in vitro* antimalarial efficacy against chloroquine sensitive (MRC-02) as well as chloroquine resistant (RKL9) strains of *P. falciparum* by serial tube dilution method. Compound 3-(4-hydroxy-phenyl)-5-(3,4,5-trimethoxyphenyl)-4,5-dihydro-pyrazol-1-yl]-pyridin-3-yl methanone (**34**) was most active one among all synthesized derivatives ((IC $_{50} = 0.0425 \mu M$). The presence of hydroxyl and methoxy moities proved the importance in enhancing antimalarial activity [46].

Hydrazide as antitubercular agents

Telvekar *et al.* reported N-benzylidene benzofuran-3-carbohydrazide derivatives and evaluated their *in vitro* antituberculosis (anti-TB) activity against M. *tuberculosis* by micro titer plate method. The results of antimicrobial activity showed compound (E) -N-benzylidenebenzofuran-3-carbohydrazide (35) was most potent among all synthesized derivatives (MIC = 2 μ g/ml). The SAR study indicated that the presence of benzofuran ring was significant in improving the antimycobacterial activity of synthesized hydrazide derivatives. [47]

Pathak *et al.* synthesized 3-(4-chlorophenyl)-4-substituted pyrazole hydrazides and evaluated their *in vitro* antitubercular activity against *M. tuberculosis* using the BACTEC 460 radiometric system. Compound *N*-(2*R*, 3*R*)-3-chloro-2-(2-hydroxyphenyl)-4-oxoazetidin-1-yl)-3-(4-chlorophenyl)-1*H*-pyrazole-5-carboxamide (**36**) was found to be most effective having MIC value of 0.35 μ g/ml. The highest antitubercular activity of compound (**36**) may be due to presence of hydrogen bond forming hydroxyl group and electron withdrawing chloro group [30].

Jordao *et al.* synthesized *N*-benzylidene benzofuran-3-carbohydrazide derivatives and evaluated their antimycobacterial activities against *M. tuberculosis* using broth macro dilution method. Compound (*E*)-5-methyl-*N*'- ((5-nitrofuran-2-yl)methylene)-1-(phenylamino)-1*H*-1,2,3-triazole-4-carbohydrazide (37) was most active one among all synthesized derivatives (MIC = $2.5 \mu g/ml$). The presence of the furyl ring with electro-negative NO₂ group was responsible for improving antitubercular activity of synthesized hydrazide derivatives [48].

Thomas *et al.* also reported their *in vitro* antimycobacterial activity against *M. tuberculosis*, *M. smegmatis* and *M. fortuitum* by Resazurin assay method. The results of activity showed compound 4-hydroxy-8-trifluoromethyl-quinoline-3-carboxylicacid(5-methyl-3*H*-imidazol-4-ylmethylene)hydrazide (38) was most active one against *M. tuberculosis*, among all synthesized derivatives (MIC = $0.625 \mu g/ml$) and this may be due to the incorporation of lipophilic indole and 8- trifluoromethyl-quinoline hydrazone moieties in synthesized derivatives [35].

Pavan *et al.* synthesized thiosemicarbazones, semicarbazones, dithiocarbazates and their hydrazide/hydrazones derivatives and evaluated their antitubercular activities against *M. tuberculosis* by Resazurin Microtiter Assay. 2-(Di(pyridin-2-yl)methylene)-*N*-phenyl hydrazine carbothioamide (39) was found to be most potent among all synthesized derivatives (MIC = $0.78 \mu g/ml$). The present antimicrobial results revealed the importance of pyridine ring in contributing antitubercular activity of synthesized derivatives [7].

Shahar *et al.* reported 1,3,4-oxadiazole derivatives and evaluated their *in vitro* antimycobacterial activity against *M. tuberculosis* using BACTEC 460 radiometric system. 2-(2-naphthyloxymethyl)-5-phenoxymethyl-1,3,4-oxadiazole (**40**) was found to be most effective having MIC value 6.25 μ g/ml. The presence of hydrophobic β -naphthyloxy methyl group at position 2 and a phenoxy methyl group at position 5 of 1,3,4-oxadiazole lead to increament in antitubercular activity as compared to other synthesized derivatives [49].

Hydrazide as antiamoebic agents

Siddiqui *et al.* synthesized hybrids of hydrazones and benzimidazole derivatives and evaluated *in vitro* antiamoebic activity against *E. histolytica* by micro dilution method. Compound N'-(4-methylbenzylidene)isonicotinohydrazide (41) was most active among synthesized derivatives (IC₅₀ = 0.131 μ M). Further, analysis of antiamoebic results showed the importance of p-methyl substituted phenyl ring in improving activity of synthesized hydrazide derivative (41) against *E. histolytic* [50].

Wani *et al.* reported synthesis of pyrazoline derivatives by cyclization of chalcones with 2-[5-(4-methoxyphenyl)-1*H*-tetrazol-1-yl]acetohydrazide and evaluated *in vitro* for antiamoebic activity against *E. histolytica* by micro dilution method. The results of antiamoebic activity showed that compound 1-(3,5-di-p-tolyl-4,5-dihydro-1*H*-pyrazol-1-yl)-2-(5-(4-methoxy phenyl) -1*H*-tetrazol-1-yl)ethanone (42) was most potent among all synthesized derivatives (IC₅₀ = 0.86 μ M). SAR results revealed that presence of electron releasing groups (OCH₃, CH₃) on phenyl ring was important for antiamoebic acitiy [51].

N, N, O N, N, O N-N O 42

CONCLUSION

Reported data showed that hydrazide derivatives have a good antibacterial, antifungal, antimycobacterial, antimalarial and antiamoebic potential. Different studies indicated that hydrazide-hydrazone derivatives with electron withdrawing groups (*viz.* NO₂, Cl) and electron donating groups (*viz.* OCH₃, N(CH₃)) have significant role in antimicrobial and antimalarial activities. Further, derivatization of hydrazides to nitrogen containing heterocyclic rings (*viz.* pyrazole, oxadiazole, azetidin-2-ones *etc.*) also improved the antimicrobial activities. In case of antitubercular activity presence of azetidin-2-ones, pyrazole, pyridine, nitro substituted furan moieties and trifluoromethyl group enhanced the activity. Presence of electron donating methyl and methoxy groups also improved the antiamoebic activity of hydrazide-hydazone derivatives.

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