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**Research Article** 

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## Investigation of drying kinetics of finger millet in fluidized bed dryer

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## ABSTRACT

Finger millet (Ragi) is a highly nutritious cereal, rich in calcium, proteins, iron and other minerals. Drying technique is usually employed to preserve Ragi for a longer time. In this work, the drying process was carried out in laboratory scale fluidised bed dryer and the drying kinetics was studied. The experimental data were made to fit with fifteen thin layer models and the model with best fit was chosen based on the statistical analysis values of  $R^2$ , RMSE and SSE. The best fit models were Modified Henderson and pabis model, Midilli model and Modified Page model II with maximum  $R^2$  values and minimum RMSE, SSE values. The experiments were also carried out to study the effect of temperature and velocity of air, initial moisture content and loading of ragi on drying time. By using fick's diffusion equation the effective diffusivity values were determined and they varied between 3.26 x 10<sup>-10</sup> and 1.29 x 10<sup>-9</sup> m<sup>2</sup>/minute. The activation energy value varied between 19.318 and 59.78 kJ/mol. The heat and mass transfer co-efficient for a single particle of Ragi was also calculated.

Key words: Finger millet, fluidised bed dryer, statistical analysis, thin layer models

## **INTRODUCTION**

Finger millet (Ragi) is a hardy crop that grows in high altitudes and withstands harsh weather conditions. In India, Karnataka is the largest producer contributing 58% in annual production. Ragi is a highly nutritious cereal loaded with calcium, proteins, iron and other minerals. It helps in reducing weight and used to treat various diseases like Brittle bones, Osteoporosis, Anaemia, Diabetes etc., In order to preserve Ragi and store it for a longer time, it needs to be dried.

Drying is employed in agro based industries for moisture removal in the presence of heat thus preserves materials by avoiding microbial growth and undesirable chemical reactions. It results in reduction of packaging, storage and transportation costs [2, 16, 21]. Fluidized bed drying is an efficient drying method due to good mixing, high heat and mass transfer coefficients and also it exhibits shorter drying time [9].

Various agricultural products like canola [8,17], ragi [10,22], green peas [7], potatoes [13], shelled corn [15], pepper corn [16], coconut [18], sweet potatoes [19] etc., and with other products like ammonium chloride [21], lignite [4], bovine intestine [11], baker's yeast [12], coated sodium per carbonate particles [1] etc., have been dried in Fluidised bed dryer and studied. Although work has been reported with ragi [10, 22], the modelling work is not yet done.

The main objective of this study is to determine the model which successfully describes the drying kinetics of ragi among fifteen thin layer models based on statistical analysis and to investigate the effect of temperature, velocity, load and initial moisture content on drying time. In addition to this, effective moisture diffusivity and activation energy is determined. Further heat and mass transfer co-efficient is also calculated.

## **EXPERIMENTAL SECTION**

## 2.1. RAW MATERIAL

Finger millet (*Eleusine coracana*) was used as the material for testing and the characteristic values are listed in table 1.

#### 2.2. EXPERIMENTAL SET UP AND PROCEDURE

The set up consist of a Glass column, the conical portion of which was filled with fluidizing material. The material was supported on the screen mesh held between two flanges. Air from the Blower was heated in the heater box and passed through the column. Orifice with differential manometer was provided to measure the air flow rate. The flow rate can be adjusted by needle valve provided for air supply to the column. Sensors were given at different positions to measure the temperature at different points. The experimental set up and schematic representation is shown in fig.1, 2.

Sample of about 150g of Ragi with the initial moisture content of 18% was loaded into the column. The set point was fixed as  $40^{\circ}$ C for temperature and 3 m/s for velocity. At regular time interval of 10 minutes sample was collected. 2g of sample was weighed, packed and dried in the hot air oven at  $105^{\circ}$ C for 24 hours to determine the moisture content. The experiments were repeated for various operating conditions as listed in table 2.

### 2.3. FLUIDISATION BEHAVIOUR

Prediction of minimum fluidisation velocity is important because the velocity should be maintained above this. Ergun equation was widely used for the determination of minimum fluidisation velocity for spherical particles [12] and it is given by

$$\frac{1.75\rho u_{mf}^2(1-\varepsilon)}{\phi d_p \varepsilon^3} + \frac{150\mu u_{mf}(1-\varepsilon)^2}{\phi^2 d_p^2 \varepsilon^3} - \rho g(1-\varepsilon) = 0$$
(1)

The terminal settling velocity is calculated using Haider and Levespiel correlation

$$u_{t}^{*} = u_{t} \left[ \frac{\rho^{2}}{\mu(\rho_{p} - u_{t}^{*}g)} \right]^{1/3}$$
(2)  
$$u_{t}^{*} = \left[ \frac{18}{(d_{p}^{*})^{2}} + \frac{2.335 - 1.744\varphi}{(d_{p}^{*})^{0.5}} \right]^{-1}$$
(3)

The experimental minimum fluidisation velocity lies between the theoretical minimum fluidisation velocity and terminal settling velocity so that the particle will be under fluidisation. Prediction of fluidisation behaviour is important and it was done by using Walli's model. The expression is given by

$$\frac{1.79}{n} \left(\frac{gd_{p}}{u_{t}^{2}}\right)^{0.5} \left(\frac{\rho_{p} - \rho_{f}}{\rho_{p}}\right)^{0.5} \left(\frac{\epsilon_{st}^{1-n}}{(1 - \epsilon_{st})^{0.5}}\right) - 1 = V_{e}$$
(4)

The fluidisation regime for corresponding values of Walli's factor  $\left(V_{e}\right)$  is

 $V_e > 0$  Homogenous regime

 $V_e < 0$  Bubbling regime

## 2.4. EFFECT OF DRYING CONDITIONS

To optimise the operating conditions for drying, experiments were conducted to study the effect of temperature and velocity of inlet air, load and initial moisture content of ragi on drying time.

### 2.5. MATHEMATICAL MODELLING OF FINGER MILLET

The experimental data obtained were fitted with fifteen models [1] to determine the model which successfully explain and predict the drying behaviour of ragi. These models uses moisture ratio and it is represented as

 $MR = \frac{X_t - X_e}{X_i - X_e}$ 

(5)

The models and their equations are listed in table 3. The modelling work was done by non-linear regression analysis using least square algorithm in MATLAB software. The co-efficient of determination ( $R^2$ ) was one of the main criteria in choosing the best fit model. In addition to this, the goodness of fit was evaluated by Root mean square error (RMSE) and Sum of squares error (SSE). For best fit model the value of  $R^2$  should be higher and value of RMSE,SSE should be lower.

## 2.6. EFFECTIVE MOISTURE DIFFUSIVITY

Fick's diffusion equation was widely used for the determination of effective moisture diffusivity and is given by

(6)

(7)

$$MR = \frac{6}{\pi^2} exp\left(\frac{-\pi^2 D_{eff} t}{r^2}\right)$$

The effective moisture diffusivity was obtained by plotting graph between ln(MR) and time.

The Activation energy can be computed using Arrhenius type equation and is given by

$$D_{eff} = D_0 \exp\left(\frac{-E_a}{RT}\right)$$

#### 2.7. HEAT AND MASS TRANSFER CO-EFFICIENT

The heat and mass transfer co-efficients are required for ideal dryer design. The heat and mass transfer co-efficient in a bed for a single particle is given by the correlations [24].

$$Nu = 2 + 1.8 \operatorname{Re}_{p}^{1/2} \operatorname{Pr}^{1/3}$$

$$Sh = 2 + 1.8 \operatorname{Re}_{p}^{1/2} \operatorname{Sc}^{1/3}$$
(8)
(9)

#### **RESULTS AND DISCUSSION**

#### **3.1. EFFECT OF OPERATING CONDITIONS ON DRYING TIME**

Fig.3 show that the drying time decreases as the inlet air temperature increases. It's due to high thermal input to the system. The rate of drying increases as inlet air temperature increases because of the intensity of heat. Fig.4 show that with the increase in velocity of air the drying time decreases and the rate of drying increases. The effect of load on drying time is unpredictable and in this case increase in load results in decrease in drying time due to variations in contacting pattern between the sample and air as shown in Fig.5. Fig.6 shows that increase in initial moisture content of ragi, the total moisture available for removal will be more so the drying time increases.

The falling rate period alone was alone observed in the rate of drying curve as shown in Fig.7. It was the typical characteristic of agricultural products as reported earlier and it symbolise that internal mass diffusion governs the process.

## **3.2 MODELLING OF DRYING KINETICS**

Modelling work for ragi was done by non-linear regression analysis using least square algorithm in MATLAB for various temperatures and velocities. Statistical results are summarised in table 4 for fifteen thin layer drying models for various conditions. Based on statistical results, Modified Henderson and Pabis model, Midilli model and Modified Page model II contains maximum values of  $R^2$  and minimum values of RMSE, SSE. So these models fit with the experimental data and successfully describe drying kinetics of ragi. The model constants for the best fit models are listed in table 5.The comparison of statistical results for the best fit models is shown in Fig.8.

#### **3.3. EFFECTIVE MOISTURE DIFFUSIVITY**

Temperature and velocity have significant effect on effective diffusivity i.e Effective moisture diffusivity increases with increase in temperature and velocity of air. The values obtained were within the range as previously reported for food materials and are tabulated in table 6.1. The values of activation energy ( $E_a$ ) and the pre-exponential factor ( $D_0$ ) values for various velocities are tabulated in table 6.2.

## **3.4. TRANSFER CO-EFFICIENT**

The heat and mass transfer co-efficient for a single spherical particle in air at various temperatures and velocities were calculated and are listed in table 7.1 and 7.2. The heat transfer co-efficient increases with increase in inlet air velocity and the mass transfer co-efficient increases with increase in inlet air temperature and velocity.

#### Table 1: Characteristics of Finger millet

Parameters	Characteristic value	
Shape of seed	Spherical	
Mean particle diameter in mm	1.071	
Density in kg/m <sup>3</sup>	1474.926	
Gas hold up	0.348	
Minimum fluidisation velocity in m/s	0.283	
Terminal settling velocity in m/s	5.6	
Fluidisation regime	Bubbling regime	

#### Table 2: Experimental conditions

Parameters	Characteristic value	
Temperature in <sup>0</sup> C	40, 50, 60	
Velocity in m/s	2.1, 3, 3.4	
Initial Moisture content	18%, 23%, 25%	
(kg of moisture/kg of dry solid)		
Load in g	150, 200, 250	

#### Table 3: Thin layer drying models used for the modelling of Ragi

No.	Model	Equation
1	Lewis or Exponential model	MR = exp(-kt)
2	Page model	$MR = exp(-kt^n)$
3	Henderson and Pabis model	$MR = a \exp(-kt)$
4	Modified Page model I	$MR = \exp(-(kt)^n)$
5	Wang and Singh model	$MR = 1 + at + bt^2$
6	Logarithmic model	MR = aexp(-kt + c)
7	Two term model	$MR = aexp(-k_0t) + b.exp(-k_1t)$
8	Two term exponential model	MR = aexp(-kt) + (1-a)exp(-kat)
9	Diffusion approach model	MR = aexp(-kt) + (1-a)exp(-kbt)
10	Verma et al model	MR = aexp(-kt) + (1-a)exp(-gt)
11	Modified Henderson and Pabis model	MR = aexp(-kt) + b.exp(-gt) + c.exp(-ht)
12	Midilli model	$MR = aexp(-kt^n) + bt$
13	Modified Page model II	$\mathbf{MR} = \exp\left(-k\left(\frac{t}{L^2}\right)^n\right)$
14	Thomson model	$t = a.\ln(MR) + b.\ln(MR)^2$
15	Simplified Fick's diffusion equation	$MR = a.exp\left[-c\left(\frac{t}{L^2}\right)\right]$

M. no	T in <sup>0</sup> C		0.1			v in m/s			2.4	
		<b>D</b> <sup>2</sup>	2.1	005	<b>D</b> <sup>2</sup>	3	005	<b>D</b> <sup>2</sup>	3.4	
		R <sup>2</sup>	RMSE	SSE	R <sup>2</sup>	RMSE	SSE	R <sup>2</sup>	RMSE	SSE
1	40	0.9264	0.0307	0.0094	0.8342	0.0420	0.0176	0.2609	0.0790	0.062
	50	0.7749	0.0642	0.0413	0.5890	0.0829	0.0688	0.6835	0.0884	0.078
	60	0.8340	0.0719	0.0517	0.6235	0.0979	0.0959	0.8682	0.0776	0.060
2	40	0.9583	0.0243	0.0053	0.9726	0.0180	0.0029	0.9902	0.0096	0.000
	50	0.9941	0.0110	0.0011	0.9726	0.0226	0.0046	0.9971	0.0090	0.000
	60	0.9829	0.0243	0.0053	0.9957	0.0104	0.0010	0.9963	0.0138	0.001
3	40	0.9394	0.0293	0.0077	0.9048	0.0335	0.0101	0.6973	0.0533	0.025
	50	0.8910	0.0471	0.0200	0.7944	0.0618	0.0344	0.8503	0.0641	0.037
	60	0.9074	0.0566	0.0289	0.8174	0.0719	0.0465	0.9231	0.0624	0.035
4	40	0.9602	0.0228	0.0057	0.9693	0.0169	0.0032	0.9606	0.0032	0.000
	50	0.9880	0.3307	1.2030	0.9242	0.0216	0.0051	0.9933	0.0082	0.000
	60	0.9719	0.0327	0.0118	0.9889	0.0096	0.0010	0.9915	0.0126	0.001
5	40	0.9765	0.0183	0.0030	0.9891	0.0114	0.0012	0.8238	0.0107	0.014
	50	0.9565	0.0298	0.0080	0.8879	0.0456	0.0188	0.8880	0.0553	0.027
	60	0.9489	0.0421	0.0159	0.8707	0.0605	0.0329	0.9299	0.0566	0.028
6	40	0.9691	0.0222	0.0039	0.9870	0.0131	0.0014	0.9804	0.0129	0.001
	50	0.9931	0.0126	0.0013	0.9842	0.0182	0.0026	0.9774	0.0264	0.005
	60	0.9784	0.0290	0.0067	0.9822	0.0238	0.0045	0.9886	0.0255	0.005
7	40	0.9866	0.0156	0.0017	0.9909	0.0118	0.0010	0.9932	0.0090	0.000
	50	0.9966	0.0094	0.0006	0.9895	0.0159	0.0018	0.9946	0.0138	0.001
	60	0.9816	0.0286	0.0057	0.9960	0.0120	0.0010	0.9964	0.0154	0.001
8	40	0.8841	0.0406	0.0148	0.9236	0.0300	0.0081	0.5285	0.0665	0.039
	50	0.8937	0.0465	0.0195	0.7636	0.0663	0.0395	0.8287	0.8097	0.068
	60	0.9213	0.0522	0.0245	0.7865	0.0777	0.0544	0.9442	0.0532	0.025
9	40	0.9729	0.0208	0.0035	0.9901	0.0115	0.0010	0.9932	0.0085	0.000
	50	0.9966	0.0088	0.0006	0.9894	0.0149	0.0018	0.9946	0.0130	0.001
	60	0.9815	0.0268	0.0058	0.9960	0.0113	0.0010	0.9964	0.0144	0.001
10	40	0.9864	0.0147	0.0017	0.9901	0.0115	0.0010	0.9932	0.0085	0.000
	50	0.9966	0.0088	0.0006	0.9894	0.0149	0.0018	0.9946	0.0129	0.001
	60	0.9815	0.0268	0.0058	0.9960	0.0113	0.0010	0.9964	0.0144	0.001
11	40	0.9788	0.0233	0.0027	0.9935	0.0117	0.0006	0.9974	0.0066	0.000
	50	0.9983	0.0078	0.0003	0.9906	0.0177	0.0016	0.9977	0.0107	0.000
	60	0.9876	0.0278	0.0039	0.9981	0.0098	0.0005	0.9975	0.0152	0.001
12	40	0.9785	0.0198	0.0027	0.9903	0.0121	0.0010	0.9970	0.0060	0.000
	50	0.9982	0.0068	0.0003	0.9826	0.0204	0.0029	0.9973	0.0097	0.000
	60	0.9849	0.0259	0.0047	0.9980	0.0086	0.0005	0.9971	0.0137	0.001
13	40	0.9583	0.0258	0.0053	0.9726	0.0191	0.0029	0.9902	0.0102	0.000
	50	0.9941	0.1164	0.0011	0.9726	0.0240	0.0046	0.9971	0.0095	0.000
	60	0.9829	0.0258	0.0053	0.9962	0.0110	0.0010	0.9963	0.0146	0.001
14	40	0.9206	9.8520	873.6	0.8694	12.6400	1437	0.6516	20.6400	3832
	50	0.8495	13.5600	1655.0	0.7527	17.3900	2721	0.8111	15.2000	2078
	60	0.8663	12.7800	1471.0	0.7759	16.5500	2465	0.8904	11.5700	1205
15	40	0.9394	0.0311	0.0077	0.9048	0.0356	0.0101	0.6973	0.0566	0.025
	50	0.8910	0.0500	0.0200	0.7944	0.0656	0.0344	0.8503	0.0680	0.037
	60	0.9074	0.0601	0.0289	0.8174	0.0762	0.0465	0.9231	0.0662	0.035

Table 4: Statistical analysis results for fifteen models for various operating conditions

T in <sup>0</sup> C	v in m/s	Μ	odified He	nderson an	d Pabis m	odel consta	nts
		а	b	с	g	h	k
40	2.1	0.0425	0.9689	-0.0116	0.0074	0.9896	-0.0166
	3	0.0184	0.9492	0.0323	0.0063	1.5590	-0.0236
	3.4	0.1005	0.2000	0.6995	0.0409	0.0002	1.8120
50	2.1	0.1990	0.6622	0.2188	0.0119	-0.0051	0.1496
	3	-0.8482	1.1490	0.6986	0.1145	0.0019	0.1384
	3.4	0.1629	0.3491	0.4879	0.0258	0.0010	0.4828
60	2.1	0.8579	0.1272	0.0143	0.7461	-0.0239	0.0109
	3	0.2973	0.1412	0.5615	1.6400	0.0021	0.0441
	3.4	0.2467	0.7313	0.0219	0.0130	-0.0153	0.1366

T in <sup>0</sup> C	v in m/s	Ν	lidilli mod	lel constan	nts
		а	b	k	Ν
40	2.1	0.9894	0.0044	0.0049	1.2260
	3	0.9952	0.0037	0.0116	0.9951
	3.4	1.0000	0.0015	0.0674	0.0476
50	2.1	0.9998	0.0017	0.0378	0.6979
	3	1.0030	0.0022	0.0517	0.6439
	3.4	0.9997	0.0004	0.0926	0.4835
60	2.1	0.9982	0.0012	0.0473	0.6980
	3	1.0000	0.0010	0.0960	0.5125
	3.4	1.0000	0.0007	0.0720	0.6554
T in <sup>0</sup> C	v in m/s	Modifi	ied Page n	nodel II co	nstants
		k		1	n
40	2.1	0.728	2 14.3	3200 (	).7625
	3	0.030	1 1.0	870 (	).5832
	3.4	0.059	8 0.3	773 (	).2846
50	2.1	0.068	0 1.1	700 (	0.5074
	3	0.618	7 11.0	)200 (	0.4078
	3.4	0.791	5 10.3	3100 (	).4355
60	2.1	0.940	6 10.6	5900 (	).5605
	3	0.111	9 0.8	763 (	0.4030
	3.4	0.145	7 15	540 (	).5757

Table 6.1: Effective diffusivity at various temperatures and velocities

2.1	3	3.4
5.71E-10	4.19E-10	3.26E-10
6.06E-10	5.36E-10	7.46E-10
8.95E-10	7.56E-10	1.29E-09
	5.71E-10 6.06E-10	5.71E-10 4.19E-10 6.06E-10 5.36E-10

Effective diffusivity in m<sup>2</sup>/minute, Temperature in <sup>0</sup>C and velocity in m/s

Table 6.2: Activation energy at different velocities

velocity in m/s	E <sub>a</sub> in kJ/mol	D <sub>0</sub> in m <sup>2</sup> /s
2.1	19318.4104	9.053E-07
3	25437.51	7.246E-06
3.4	59780.9856	3.2152336

#### Table 7.1: Heat transfer co-efficient H at various temperatures and velocity

Temperature/Velocity	2.1	3	3.4		
40	524.9837039	617.4279224	653.9784814		
50	524.4913457	616.0452243	652.995894		
60	523.8416593	615.5525164	651.8131193		
Heat transfer co-efficient in $W/m^2K$ ; Temperature in <sup>0</sup> C and velocity in m/s					

Table 7.2: Mass transfer co-efficient  $k_d$  at various temperature and velocity

2.1	3	3.4
0.462674844	0.544140909	0.57635093
0.477287597	0.561090255	0.594224118
0.492171048	0.578342731	0.612413257
	0.462674844 0.477287597	0.462674844 0.544140909 0.477287597 0.561090255

Mass transfer co-efficient in m/s; Temperature in  ${}^{0}C$  and velocity in m/s



Fig. 1 Experimental set up of fluidised bed dryer 1-Blower, 2- Preheater, 3- Drying chamber, 4- Cyclone separator, 5- Water manometer, 6- Temperature indicator

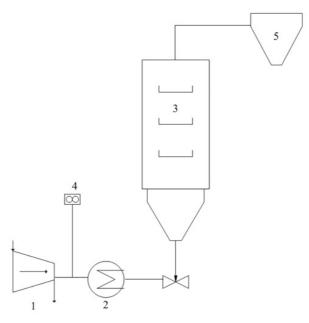
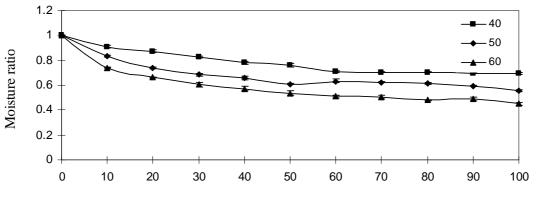
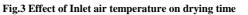
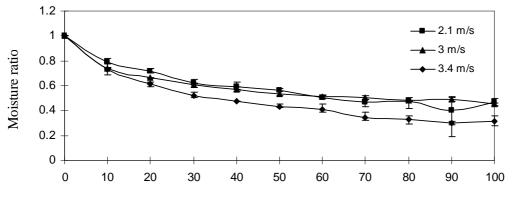


Fig. 2 Schematic representation of Fluidised bed dryer 1-Blower, 2-Preheater, 3-Drying chamber, 4-Water manometer, 5-Cyclone separator



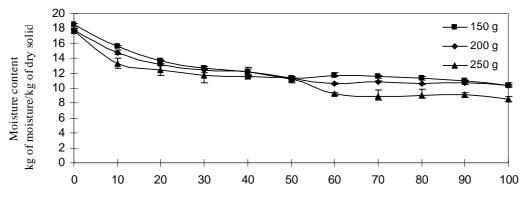
Time in minutes





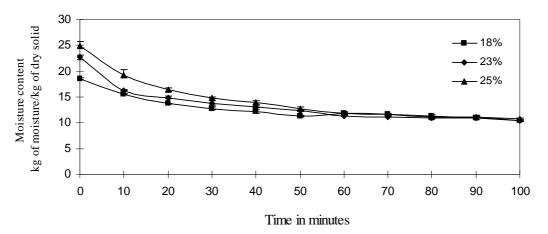
Time in minutes

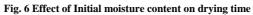
Fig.4 Effect of inlet air velocity on drying time



Time in minutes

Fig.5 Effect of load on drying time





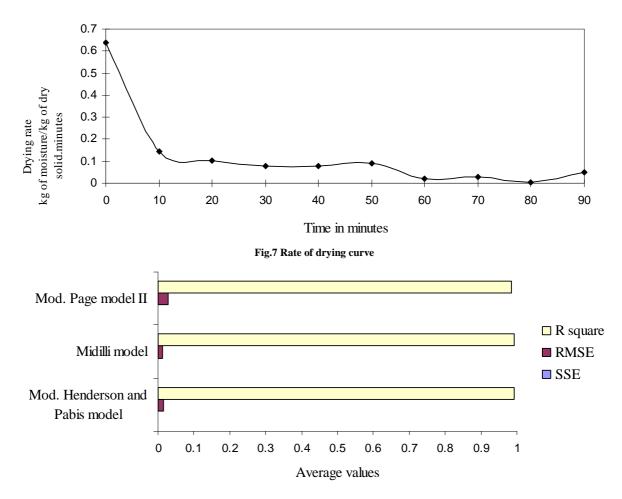


Fig. 8 Comparison of Statistical results for best fit models

## CONCLUSION

The drying time of finger millet decreases with increase in inlet air temperature, velocity and load of sample, decrease in initial moisture content of sample. The falling rate period alone was present in the rate of drying curve which indicates internal mass diffusion governs the process. Among fifteen widely used thin layer models, Modified Henderson and Pabis model, Midilli model and Modified Page model II were found to be best fit models with maximum  $R^2$  and minimum RMSE,SSE values and they successfully describes the drying kinetics.

The effective moisture diffusivity varies between  $3.26 \times 10^{-10}$  and  $1.29 \times 10^{-9}$  m<sup>2</sup>/minutre. The activation energy ranges between 19.318 and 59.78 kJ/mol. The value of heat transfer co-efficient is in the range of 523.84 and 653.97W/m<sup>2</sup>K. The mass transfer co-efficient is in the range of 0.462 and 0.612m/s.

#### NOMENCLATURE

NOMENCLATORE	
a,b,c,k <sub>0</sub> ,k <sub>1</sub> ,k,n,g,h,L	Constants of the drying models
$D_{eff}$	Effective moisture diffusivity (m <sup>2</sup> /minute)
d <sub>p</sub>	Particle diameter (m)
$\dot{\mathbf{D}_{o}}$	Pre-exponential factor (m <sup>2</sup> /minute)
$\mathbf{E}_{\mathbf{a}}$	Activation energy (kJ/mol)
g	Acceleration due to gravity(m/s <sup>2</sup> )
H	Heat transfer co-efficient of a single sphere in gas $(W/m^2K)$
K <sub>d</sub>	Mass transfer co-efficient of a single sphere in gas (m/s)
MR	Moisture Ratio
n	Richard zaki co-efficient
Nu	Nusselt number
Pr	Prandtl number
r	Radius of particle (m)
R	Universal gas constant (8.314kJ/kmol.K)
Re	Reynolds number
Sc	Schmidt number
Sh	Sherwood number
Т	Temperature (K)
t	Time (minutes)
$\mathrm{U}_{\mathrm{mf}}$	Minimum fluidisation velocity (m/s)
Ut	Terminal settling velocity (m/s)
X <sub>t</sub>	Moisture content at time t (kg of moisture/kg of dry solid)
X <sub>i</sub>	Initial moisture content (kg of moisture/kg of dry solid)
Xe	Equilibrium moisture content (kg of moisture/kg of dry solid)
Symbols	
Q	Density of gas (kg/m <sup>3</sup> )
ε	Gas phase hold up
φ	Sphericity
μ	Viscosity of gas (kg/m.s)
· ρ	Density difference between particle and gas
٢	
$\Delta X$	Difference in moisture content
$\Delta t$	Difference in time
<u> </u>	

#### REFERENCES

[1] S Hematian ; F Hormozi. Powder Technology 2015 269, 30–37.

[2] H Darvishi ; MH Khoshtaghaza ; S Minaei. Journal of the Saudi Society of Agricultural Science 2014.

[3] M Khamforoush; SM Mirfatah; T Hatami. *Ije Transactions b: applications* **2014** vol. 27, no. 5, 667.

[4] JH Park ; C Ha Lee ; YC Park ; D Shun ; Dal-Hee Bae ; J Park. Drying technology **2014** 32, 268.

[5] G Srinivas ; Y Pydi settee. Heat and mass transfer 2014.

[6] R Amiri Chayjan; Q Abedi; A Sabziparvar. Journal of food science & technology 2013 50(4), 667.

[7] B Honarvar. Iranian Journal of chemical engineering 2013 vol. 32, no. 1.

[8] N Malekjani ; SM Jafari ; MH Rahmati ; EE Zadeh ; H Mirzaee. Journal of Food Engineering 2013 9(4), 375.

[9] SDF Mihindukulasuriya ; HPW Jayasuriya. Agric Eng Int: CIGR Journal 2013 Vol. 15, No.1, 154.

[10] SP Shingare ; BN Thorat. Drying Technology 2013 31,507.

[11] W Senadeeraa; O Alves-Filhob; T Eikevikb. Food and Bioproducts Processing 2013 9 1, 549.

[12] H Akbaria ; K Karimia ; Magnus Lundin ; MJ Taherzadeh Food and Bioproducts Processing 2012 90, 52.

[13] S Balasaheb ; GP Sharma ; SP Sonawane ; RC Verma. Journal of food science & technology 2012 49(5), 608.

[14] R Amiri Chayjan ; J Amiri Parian ; M Esna-Ashari. Spanish journal of agricultural research 2011 9(1), 28.

[15] L Momenzadeha; A Zomorodiana; D Mowlab. Food and Bioproducts Processing 2011 8 9, 15.

[16] P Promvonge ; A Boonloi ; M Pimsarn ; C Thianpong. International Communications in Heat and Mass Transfer 2011 38, 1239.

[17] H Gazor ; A Mohsenimanesh. Czech Journal of Food Science 2010 Vol. 28, 531.

[18] T Madhiyanon ; A Phila ; S Soponronnarit. *Applied thermal engineering* **2009** 29, 2849.

[19] MS Hatamipour ; H Hadji Kazemi ; A Nooralivand ; A Nozarpoor. *food and bioproducts processing* **2007** part c.

[20] SM Tasirin; SK Kamarudin; K Jaafar; KF Lee. Lee Journal of Food Engineering 2007 79, 695.

[21] R Kumaresan ; R Virudhagiri. Indian journal of chemical technology 2006 vol 13, 440.

[23] RE Treybal. Mass Transfer Operations third Edition, McGraw-Hill International editions 1955.

[24] D Kunii ; O Levenspiel. Fluidisation Engineering second edition, Butterworth-Heinemann series in Chemical Engineering **1991**.

<sup>[22]</sup> P Sivashanmugam ; S Sundaram. Powder Technology 2007 107, 256.