# Journal of Chemical and Pharmaceutical Research



J. Chem. Pharm. Res., 2010, 2(2): 300-303

# An environmentally benign organic solvent free approach for synthesis of new Schiff bases and evaluation of antibacterial activity

Archana Y. Vibhute, Subhash B. Junne, Vasant M. Gurav and Yeshwant B. Vibhute\*

Organic Synthesis Laboratory, P.G. Department of Chemistry, Yeshwant Mahavidylaya, Nanded. (M.S.) India

#### Abstract

Eight new heterocyclic moiety containing Schiff bases have been synthesized by the condensation of aromatic amines with substituted benzaldehyde under organic solvent free condition efficiently in the presence of water. The Schiff bases were obtained in good yields and were easily isolated by filtration. Their structures were confirmed by IR, <sup>1</sup>HNMR and elemental analysis. Most of the Schiff bases have showed potent antibacterial activity.

Key Words: Schiff bases, Synthesis, Organic solvent free, Antibacterial activity.

#### Introduction

The Schiff bases constitute one of the most active class of the compounds possessing diversified biological activity such as antitubercular [1],anticancer [2], antibacterial [3-10], antifungal [10], analgesic [11], CNS depressant [11], anti-inflammatory [12], anticonvulsant [13], insecticidal [14], plant growth inhibitors [15]. Schiff bases are used as starting material for the synthesis of various bioactive heterocyclic compounds like 4-thiazolidinones, 2-azetidinones, benzoxazines and formazans. One of the important role of Schiff base is an intermediate in the biologically important transamination reaction. Schiff bases are used as protective agent in natural rubber [16]. Schiff bases are used as amino protective group in organic synthesis. Dabholkar and More [17] have synthesized Schiff bases under microwave irradiation. Recently Schiff bases [ 18,19] have been synthesized by condensing carbonyl compounds and amines in water suspension medium. These wide application and diverse potential biological activities of Schiff bases prompted us to synthesize new Schiff bases containing heterocyclic moiety and to as certain their microbial activity.

#### **Materials and Methods**

#### **Experimental Section**

All melting points were taken in open capillaries and are uncorrected. FT-IR spectra were recorded on perkin-Elmer-157 spectrophotometer instrument using KBr discs. <sup>1</sup>HNMR were taken on a Bruker WN-400 FTMHz NMR instrument using DMSO/CdCl<sub>3</sub> solvent and TMS as a internal standard.

#### Typical procedure for preparation of Schiff bases

A mixture of aldehydes and aromatic amines (0.01 mol) were taken in mortar. Added to it acetic acid (0.25 ml), water (5 ml.) and stirred at room temperature for 30-45 min. Reaction was monitored on T.L.C. After completion of reaction, water (25 ml) added.

Separated solid was filtered, washed with water and crystallized from ethyl alcohol. Physical and analytical data is given in Table-1.

#### Scheme - 1.

Ar-NH<sub>2</sub> + Ar`CHO 
$$\xrightarrow{H_2O}$$
 Ar-N=CH-Ar`  
(la-h)

#### **Antibacterial Activity:**

Synthesized Schiff bases were evaluated for their antibacterial activity against plant pathogen *Xanthomonas citri (Xc), Ervinia carotovara (Ec)* and animal pathogen *Escherichia coli (E.coli)* and *Bacillus subtilis (Bs)*. An activity was studied using disc diffusion method [20] by measuring diamenter of zone of inhibition in mm. The compounds were dissolved in 5% aqueous DMF at the concentration of 150 ppm and discs were soaked and incubated at 27°C for 24 hr. Ampicillin 150 ppm was used as a standard antibiotics for comparision. All the compounds tested showed good inhibitory action but compounds Ia, Ic and Ih. showed slightly more inhibitory action than standard (Table-1).

#### **Results and Discussion**

In this communication, we have prepared eight new Schiff bases under solvent free condition. Halogenosubstituted hydroxy benzaldehydes and heterocyclic amines were taken in a morter. Added to it traces of acetic acid and water to wait the reaction mixture. Reaction mixture was grinded for 30-45 min. Reaction was monitered on T.L.C. After completion of reaction, water was added and stirred. Separated solid was filtered, washed with water and crystallized from ethyl alconol. Structures of the Schiff bases were confirmed by IR, <sup>1</sup>HNMR and elemental analysis. This procedure eliminates the use of organic solvent, completes within 30-45 min. and isolation of the product is simple (Table-1).

### Conclusion

Procedure of synthesis of Schiff bases eliminates the use of organic solvent. Reaction completes within 30-45 minutes. Isolation of the product is simple. All the compounds are more or less active to tested bacteria. Compounds Ia, Ic and Ih. found slightly more inhibitory to bacteria than standard.

## Yeshwant B. Vibhute et al

#### Table 1 Analytical data, elemental analysis, spectral data and antibacterial activity of compounds (Ia-h)

Form Ar Al M.P. Yield Crystal completion Found (Calculated)	vial activity	ntimicrobial	A	Spectral analysis			nalysis	Elemental a	Time for	itu unu u				nalytical data, ele	
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	nhibition ter 12 hour.	Zone of inhi n mm) After	(i	ILINIMD (8)	IR (cm <sup>-1</sup> )		Found (Calculated)		completion of reaction				Ar <sup>1</sup>	Ar	Entry
$ \begin{bmatrix} Ia & ( \downarrow \downarrow \downarrow \downarrow Ia) & ( \downarrow \downarrow \downarrow \downarrow Ia) & (Ia) & (I$	Ec. Xc		Bs	- 'HNMR (ð)		C=N	N	X= (I,Cl,Br)	(min)						
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	29 36	25 2	23		1595,1490	1612		20.40 (20.88)	40	Colorless	86	192	но СІ он	N S	Ia
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	21 18	14 2	18		1605,1580	(NH)			40	Pale pink	73	239	НО СІ СІ	NH-	Ib
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	27 34	23 2	27	12.05 (s,1H,OH),	1615,1602	1628			40		66	188	ОМе	N S	Ic
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	20 33	21	19		1612,1500	1622			35	Yellow	90	192	HO	N S	Id
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	16 18	12	13	12.35 (s,1H,OH),	1618,1605	1630			46	Yellow	92	187	HO Br Br	N S	Ie
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	13 16	15	14	8.35 (s,1H,=CH), 12.90 (s,1H, OH), 7.45 (s, 1H, 2Ar-H),	1615,1593	(N-H)			30		65	252	НО СІ СІ		If
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	15 22	14	16	4.05 (s, 3H,OCH <sub>3</sub> ), 8.55 (s, 1H, = CH), 7.35, (s1H,=2Ar-H), 7.52 (s, 1H,6Ar-H),	1590,1582	(NH)			30	Colorless	81	250	СІ	N N	Ig
	26 32	19 2	24	8.40 (s, 1H, =CH), 12.85 (s, 1H,OH),	1608,1586	1615			35	Colorless	68	285			Ih
Ampicillin  25  22    Streptomycin	 25 33														-

#### Acknowledgements

The authors are thankful to principal, Yeshwant Mahavidyalaya, Nanded for providing laboratory facilities and the Director of IICT, Hyderabad for providing spectral analysis.

#### References

- [1] JR Marchant; DS Chothia. J. Med. Chem., 1970, 13, 335-338.
- [2] MS Singare; DB Ingle. J. Indian Chem. Soc ,. 1976, 53, 1036-1037.
- [3] AV Dobaria; JR Patil; J Padaliya; HH Parekh. Indian J. Heterocyclic chem., 2001, 11, 115-118.
- [4] SM Nair; IRA Bhattacharya. Asian Journal of Chemistry. 2009, 21, 504-510.
- [5] S Shah; R Vyas; RH Mehta. J. Indian chem. Soc., 1992, 69, 590.
- [6] J Parekh; P Inamdha; R Nair; S Balusa; S Chanda. J. Serb.chem.Soc., 2005,70, 1155-1161.
- [7] VSV Satyanarayana; P Sreevani; A Sivakumar; Vijakumar. ARKIVOC 2008, 17, 221-223.
- [8] B Sutariya; SK Raziya; S Mohan; SV Sambasiva Rao. Indian J.chem., 2007, 46B, 884-887.
- [9] S.Bairagi; A Bhosale; M.N Deodhar. E-Journal of chemistry, 2009, 6,759-762.
- [10] AP Mishra; M Soni. Metal-based drugs, 2008, II, 875410, 1-7.
- [11] JK Gupta; De. Biplab; VS Saravanan. Indian J. Chem., 2006, 45B, 2580-2582.
- [12] S Bawa; Suresh Kumar. Indian J.Chem., 2009, 48B, 142-145.
- [13] M Verma; S. N Pandeya; K N; Singh J P Stables. Acta Pharm., 2004,54,49-56.

[14] NS Kozlov; GP Korotyshova; NG Rozhkora; EI Andreeva. Vesti Akad Navuk USSRser khim. Navuk, **1986**, 2, Chem. Abstr. **1987**, 106, 155955.

[15] S Huneck; K Schreiber; H D Grimmecke. J. plant growth Regul., **1984**,3,75-84. Chem. Abstr. **1985**,102,1871.

- [16] R S George; R Joseph; KE George. Int. J. Polym Matter, 1993,23,17-26.
- [17] V V Dabholkar; H D More. Indian J. Chem., 2004, 43B, 682.
- [18] AA Jarrahpour; D Khalili. Molecules, 2006,11,59-63.
- [19] Rajeshwar Rao; Reddy; M Mohan. Indian J.Hetercyclic chem. 2003,13,69.
- [20] C H Collins. Microbiological methods, Butterworth; London 1967, 364