



## Understanding Single Molecule Adsorption on Nanobubbles at Electrode Interfaces

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### DESCRIPTION

Nano Bubbles (NBs), tiny gas-filled cavities in liquids, have been a topic of great interest in recent years. Despite their nanoscale dimensions, they have a profound influence on various physicochemical processes, such as water cleaning, material processing, and catalysis. One area of particular interest is the interaction of fluorophores with electrode-supported nanobubbles. Fluorophores are molecules that can re-emit light upon absorption, making them ideal probes for studying nanobubble properties and behaviors. In the experimental setup, an electrode is typically immersed in a solution containing fluorophores and gas is supplied to form nanobubbles on the electrode surface. These bubbles can be observed with methods like Total Internal Reflection Fluorescence Microscopy (TIRFM), which can detect the fluorescence of fluorophores adsorbed on the NB surface.

The transient adsorption behavior of single fluorophores on an electrode-supported nanobubble describes how individual fluorophore molecules interact with the nanobubble over time. This interaction typically follows a stochastic or "random" process. As fluorophores move through the solution, they may randomly encounter the NB and adsorb onto its surface. Once adsorbed, they can desorb back into the solution, also in a random manner. When a fluorophore adsorbs onto the NB surface, its proximity to the electrode can cause its fluorescence intensity to increase due to a phenomenon known as Metal-Enhanced Fluorescence (MEF). If the NB is sufficiently small, the adsorption of a single fluorophore can be detected as a step-like increase in fluorescence intensity. Similarly, when the fluorophore desorbs, a step-like decrease in fluorescence intensity is observed.

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By monitoring these step-like changes in fluorescence intensity over time, it is possible to construct a "dwell time" histogram showing the distribution of the time periods that individual fluorophores spend adsorbed on the NB. The shape of this histogram can provide information about the kinetics of the adsorption and desorption processes. The transient adsorption behavior of fluorophores on NBs is influenced by several factors. For instance, the fluorophore's size, charge, and hydrophobicity can affect its adsorption onto the NB. The presence of other molecules in the solution can also impact the adsorption behavior by competing for adsorption sites or by modifying the properties of the NB surface. Understanding the transient adsorption behavior of fluorophores on NBs has several potential applications. For instance, it can provide insights into the properties of the NB surface, which can be relevant for applications such as catalysis and materials processing. Moreover, it can shed light on the mechanisms of fluorescence enhancement phenomena such as MEF, which are useful in developing sensitive detection methods in analytical chemistry and bio imaging. The study of the transient adsorption behavior of single fluorophores on an electrode-supported nanobubble holds significant potential for various scientific and technological applications.

The ability of nanobubbles to stabilize fluorophores and increase their photostability can be utilized for environmental monitoring and pollution remediation. By tracing the interaction of fluorophores with pollutants adsorbed on the nanobubble surfaces, it's possible to monitor and assess the presence and concentration of pollutants in the environment. Investigating the transient adsorption behavior of fluorophores on nanobubbles can aid in understanding the surface characteristics, stability, and lifespan of the nanobubbles. This could lead to the development of more efficient nanobubble generation techniques and their effective application in various industries, including wastewater treatment, mineral flotation processes, and food processing. Fluorophores interactions with nanobubbles can be harnessed for advanced imaging techniques. For instance, the enhanced fluorescence due to the proximity to nanobubbles can improve the sensitivity and resolution of bioimaging techniques, providing detailed insights into biological processes at the cellular or even molecular level. This could have significant implications for early disease detection and understanding disease progression.

In conclusion, studying the transient adsorption behavior of single fluorophores on an electrode-supported nanobubble provides valuable insights into the properties and behaviors of nanobubbles. Despite the complexity and randomness of these processes, careful experimental design and data analysis can yield meaningful and useful information. These studies contribute to our fundamental understanding of nanobubbles and can guide the development of applications ranging from catalysis to bioimaging.